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Optimal Design of Drying Process of the Potatoes with Multi-Agent Reinforced Deep Learning

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ABSTRACT

Heat and mass transport through evaporation or drying processes occur in many applications such as food processing, pharmaceutical products, solar-driven vapor generation, textile design, and electronic cigarettes. In this paper, the transport of water from a fresh potato considered as a wet porous media with laminar convective dry air fluid flow governed by Darcy's law in two-dimensional is highlighted. Governing equations of mass conservation, momentum conservation, multiphase fluid flow in porous media, heat transfer, and transport of participating fluids and gases through evaporation from liquid to gaseous phase are solved simultaneously. In this model, the variable is block locations, the object function is changing water saturation inside the porous medium and the constraint is the constant mass of porous material. It shows that there is an optimal configuration for the purpose of water removal from the specimen. The results are compared with experimental and analytical methods Benchmark. Then for the purpose of configuration optimization, multi-agent reinforcement learning (MARL) is used while multiple porous blocks are considered as agents that transfer their moisture content with the environment in a real-world scenario. MARL has been tested and validated with previous conventional effective optimization simulations and its superiority proved. Our study examines and proposes an effective method for validating and testing multiagent reinforcement learning models and methods using a multiagent simulation.

KEYWORDS

Multi-agent; reinforced learning; heat transfer; mass transfer; drying; darcy flow; moisture; optimization

Nomenclature

A	Surface area of the sample, m^2
A_{spec}	Specific area, m^2
a_w	Water activity, dimensionless
Bi_t	Thermal Biot number, $\frac{h_t d}{k}$
Bi_m	Mass Biot number, $\frac{h_m d}{D}$
C	Molar concentration, mol/m^3
c_p	Heat capacity, $J/(kg \cdot K)$
c_w	Specific heat of water, $J/(kg \cdot K)$



c_{pv}	Specific heat of water vapor, J/(kg·K)
c_{pa}	Specific heat of air, J/(kg·K)
c	Mass concentration, kg/m ³
c_a	Concentration of air, kg/m ³
$c_{v,sat}$	Saturated vapor concentration, kg/m ³
D	Diffusion coefficient, Mass diffusivity, m ² /s
D_a	Air diffusion coefficient, m ² s ⁻¹
$D_{a,eff}$	Effective heat transfer coefficient of air, m ² /s
D_{cap}	Capillary diffusion coefficient due to concentration gradient, m ² /s
D_T	Capillary diffusivity due to temperature gradient, m ² /s
D_{va}	Vapor diffusion coefficient, m ² /s
d	Sample thickness, m
F	Volume force, N/m ³
Fo	Dimensionless time, Fourier number, $\frac{\alpha t}{d^2}$
h_{evap}	Latent heat of vaporization of pure water, J/kg
h_t	Convective heat transfer coefficient, W m ⁻² K ⁻¹
h_m	Surface mass transfer coefficient, m/s
H	Height, m
J	Mass flux, kg/(m ² ·s)
k	Thermal conductivity, W/m K
K	Permeability, m ²
K_c	Effective liquid permeability of the elemental slice, m ²
K_{sat}	Saturated liquid permeability (characteristic of the material structure), m ²
K_{evap}	Evaporation rate constant, 1/s
K_r	Relative liquid permeability of the elemental slice
k_{ar}	Air's relative permeability
k_{ir}	Water relative permeability
Ko	Kossovitch number, $\frac{h_{evap} X_0}{C_p \Delta T}$
L	Length, m
Lu	Luikov number, Lewis number, D/α
M	Molecular weight, kg/mol
M_d	Moisture content, dry basis, kg of water/kg of dry solids
M_v	Molecular weight of water vapor, kg/mol
M_a	Molecular weight of air, kg/mol
M_w	Moisture content, wet basis, kg of water/kg of total
m_v	Surface evaporation rate, kg/(m ² ·s)
p	Pressure, Pa
P_l	Liquid pressure (related to gas phase pressure and capillary pressure), Pa
P_c	Capillary pressure in the elemental slice (function of saturation), Pa
P_g	Gas phase total pressure
P_v	Partial pressure of vapor, Pa
P_a	Partial pressure of air, Pa
Pn	Posnov number, $\frac{\delta \Delta T}{X_0}$
P_v	Vapor pressure, Pa

q'	Heat source, W/m^3
R	Gas constant, $J/(mol \cdot K)$
R_{evap}	Evaporation rate, $kg/(m^3 \cdot s)$
RH	Relative humidity at any time, t
S	Degree of saturation of the elemental slice (related to moisture content)
S_w	Water saturation = $V_w/\phi V$, dimensionless
S_g	Gas saturation = $V_g/\phi V$, dimensionless
S_{ir}	Relative liquid saturation
t	Time, s
T	Temperature of the elemental slice at any time t , K
T_0	Initial temperature, K
T_a	Drying air temperature, K
T_{in}	Inlet temperature, K
u	Velocity, m/s
u_{in}	Inlet velocity, m/s
W	Width, m
x	Distance, m
X	Location of interface, m
V	Total volume, m^3
X	Dry basis moisture content, $kg \text{ water}/kg \text{ d.b}$
X_0	Initial dry basis moisture content, 1
X^*	Dimensionless water content, $\frac{X}{X_0}$
Z	Dimensionless component of coordination through the thickness, $\frac{z}{d}$

Greek Symbols

α	Thermal diffusion coefficient, $m^2 s^{-1}$
δ	Thermo-gradient coefficient, $kg \text{ water}/kg \text{ d.b}/K$
ϵ	Phase change coefficient
ΔT	Temperature difference, $T_{a,in} - T_0$, K
θ	Dimensionless Temperature, $\frac{T - T_0}{T_{a,in} - T_0}$
λ	Thermal conductivity, $W m^{-1} K^{-1}$
ϕ	Porosity, dimensionless
ρ	Density, $kg m^{-3}$
ρ_a	Density of air, $kg m^{-3}$
ρ_w	Density of water, kg/m^3
τ	Dimensionless time, s
μ_a	Dynamic viscosity of air, $N s m^{-2}$
ν_a	Kinematic viscosity of air, $m^2 s^{-1}$
Γ	Rate of evaporation, $kg/m^3 s$

1 Introduction

In recent years, there have been many studies to use artificial intelligence for optimization in engineering to address real-life problems, analyze existing diverse multiagent simulations and describe

simulation optimization [1]. In an Automated guided vehicles warehouse, researchers automate the path of numerous autonomous Automated guided vehicles using centralized training with decentralized execution framework-based MARL [2]. The major benefit of MARL scheme is that it does not need any model of the system. We present a new method based on deep independent Q-learning. This modified model includes more than an observation function and a common scheme regarding its policy. It also incorporates a direct sharing of information between neighboring agents as part of the observation function [3].

To examine a cache-aided network using scalable caching under the MARL language, distributed caching is modeled using a MARL framework as a fully cooperative repeated game [4]. As shown by distributed caching, caching is more likely to occur. A supervisory network separates sub-goals for each agent to achieve a global goal. Gradually gets more complex so we can learn policy in a large-scale environment. As part of our method evaluation, Do et al. [5] do the treasure hunter experiment which proves the capability of MARL over the usual genetic algorithm [6]. An extension of single agent Actor-Critic methods is proposed in this paper for a multi-agent reinforcement learning algorithm. In the algorithm, all agents have access to one another's actions and rewards, as well as a set of Q values and techniques. For the computation of Q values, we use linear programming. A mixed Nash equilibrium is reached with a smaller value, and the algorithm has the same convergence properties as Nash equilibrium [7].

As illustrated by researchers, the suggestion of phase-based control structures for the operation of multi-agents [8] (for the application of semi-batch reactors) and sustainability enhance HVAC optimal control [9] (for the application of heating, ventilation, and air-conditioning), and optimal coordinated method for large-scale multi-agent deep reinforcement learning [10] (for the application of proton exchange membrane fuel cell heatmanagement) increased the performance of control procedures and resource management. Some advantages exist in the use of a multi-agent deep reinforcement learning algorithm for optimizing building cooling water systems including cooling towers, pumps, and chillers [11]. Considering a multi-agent scenario where each agent controls a single sub-equipment defeats that barrier. As opposed to traditional methods that rely on equipment model parameters, MARL is model-independent. Furthermore, it is necessary to constantly update the model (model-based) to maintain accuracy, which is costly. MARL has applications in grid systems optimization [12].

Proximal Policy Optimization for localizing targets [13] with various intensities is modeled in the way that agents that can perform as humans do so by cooperating to find the target by expanding the scalability and adaptability of the models. The superiority of MARL over Bayesian, uniform survey, and single-agent, and Double Deep Q-Networks is proved [13]. MARL can generate optimal policies that outperform conventional rule-based policies. Resource management for energy application of behind-the-meter resources is optimized with MARL to provide essential community services [14]. Selimefendigil et al. [15] propose a machine-learning method for the optimal spacing between multiple porous moist objects in a combined heat and mass transport environment. As the surfaces do not have uniform heat and mass transfer coefficients, a neural network model was used.

Drying in a porous media has many applications in the food industry, pharmaceutical products being dried before packaging, and in paper and board manufacturing [16]. Some papers collectively address the research question of achieving environment coupling. Beyhaghi et al. [17,18] propose a simulation approach that combines the invasion-percolation algorithm for water redistribution inside the network with a finite-volume-based code for mass transfer outside the model. Dullien [19] investigates the invasion front and evaporation front during drying, highlighting the effectiveness of the disconnected cluster erosion mechanism.

Huinink et al. [20] focus on multi-level modeling which is faster than through the solid matrix in isothermal drying [21–25]. Metzger et al. [26] extend a network model for isothermal drying of capillary porous media to account for viscosity in the liquid phase specially for mono-modal and bimodal pore [26]. In porous capillary materials (for capillary pore diameter of less than 0.1 μm) such as used in commercial gels and some ceramics [27–32] liquid water transport includes molecular diffusion as the mean free path of molecule is higher than the pore size. Prat [33] has modeled drying of porous media based on percolation theory (a network of pores connected by narrow throats) which validated to experimental results. As well in porous with non-hygroscopic and non-shrinking materials, water could be ignored [34–38] while equilibrium moisture isotherms are important in hygroscopic and shrinking materials which bound water is significantly [39–44]. Yiotis et al. [45] study capillarity-driven flow through macroscopic liquid films during the isothermal drying of porous materials. Their study shows that film flow is a major transport mechanism in the drying of pores. The mixture multiphase model solves for each phase [46,47]. The gas phase is modeled as a multi-component because typically the gas phase includes air and the evaporated solvent (steam/water vapor in this case) [48]. Since the gas phase is modeled as a multi-component [49], the liquid phase needs to be a multi-component phase as well [50].

In some models [15,51], as a first step, the viscous resistance of liquid water and gas is assumed to be constant throughout drying and porous inertial resistance as well [52]. In the porous media region under multiphase interaction the Spalding Evaporation/Condensation model [53], wherein the Nusselt number and Sherwood are given by the Armenante-Kirwan correlation [54] to capture the heat [55] and mass transfer, respectively [56]. Although the vaporization latent heat at 100°C is a constant (2257477 J/kg), as a trick to simulation stability in some papers the treatment of heat of formation and to improve the stability of the simulation, we set the heat of formation of liquid water is another value [57], and heat of formation of vapor is zero [58]. A stability trick is to assign the whole heat of formation to one component that takes part in the evaporation [59]. By default, both gas and liquid phases have heat of formation defined and the values can be high [60].

In order to classify drying in porous media problems [61], some papers consider four cases based on pore size and capillary pressure. In samples with large pores, and driving force is applied pressure which needs the solving of the governing equations [62]. In the case of small pores of rehydration with only capillarity effects and no significant internal evaporation one should solve the Darcy equation through porous media [63–65].

The main goal of the present work is the optimal design of porous block in the drying process and the geometric location is the principal conclusion of a similar study [66]. This article describes how to set up a multiphase model to simulate the drying of wet paper by convective heat transfer with previous porous media techniques [67]. The purpose of the work presented here is to develop a new analysis of drying and simulation of heat and mass transfer. This would have significant benefits in the design and operation of large-scale industrial drying systems [68–70]. The purpose of this paper briefly includes a build-up of a comprehensive simulation of potato drying in a variety of configurations, experimental validation, and application of MARL optimization for industrial drying for the moist bluff body inside channel flow configuration.

2 Materials and Methods

The purpose of the model is to analyze the details of the physics of transport mechanisms involved inside and outside the material during drying. Fig. 1 shows the schematic of the problem. First, the hot surface drying of a paper and board is considered. Paper and board are complex heterogeneous porous

materials consisting of cellulose fiber, inorganic pigments water, and air. An average velocity and low ambient humidity are imposed on the incoming airflow. The properties of the porous medium, as well as the relative humidity, are configurable through parameters. To build up a mathematical model of transport by the solid, liquid, and gas streams, evaporation of water from within and outside the porous media, diffusion of water vapor driven by concentration difference, convection of water vapor and air driven by gas phase pressure difference, condensation of evaporated vapor in cooler regions of the porous media, diffusion-condensation cycles, diffusion of air, flow of liquid in the capillaries driven by capillary pressure difference or total hydraulic pressure difference, and convective heat transfer due to the movement of the liquid and gas, all occurring simultaneously. In addition to the above, paper is also a hygroscopic material, i.e., it closely interacts with the surrounding environment. As a result, the surrounding moist environment can influence the equilibrium moisture content as well as the equilibrium vapor pressure in the vapor phase in the fiber mat at any temperature. The potential driving forces that may contribute to the above physical mechanisms are schematically shown in the following Fig. 1a. The schematic of the porous media with an elemental volume and possible transport mechanisms (computational model with 3 porous moist objects) which will be used for optimization purposes is shown in Fig. 1b.

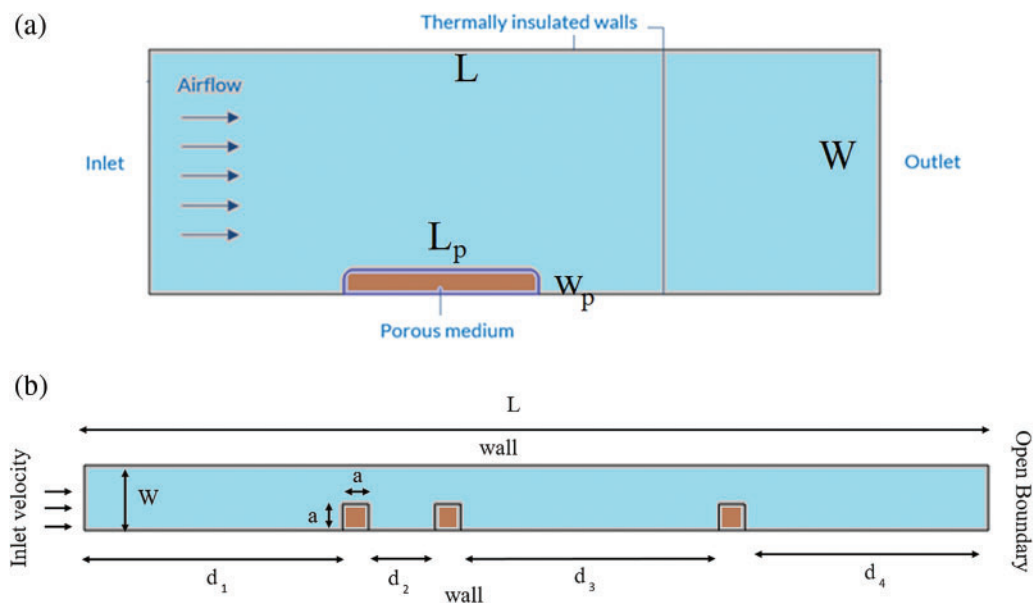


Figure 1: Schematic of the problem (a) benchmark case compared with experimental data (b) current optimization problem with geometrical parameters

2.1 Governing Equations

The development of the model equations involves deriving the equations representing each of the transport mechanisms in an elemental volume. First, we consider only the heat and mass transfer in one dimensional (thickness dimension). In the future, this can be expanded to 2 or 3 dimensions. Considering an elemental volume at any position x (in the thickness direction) within the domain of the porous media (Fig. 1), the general conservation equation can be given as (storage (or) accumulation is equal to in – out + generation). In the following derivation, only the xy -plane is considered. The governing equations of the transient process include continuity equations in the free stream are defined by air density and the superficial velocity of the fluid as

$$\frac{\partial \rho_a}{\partial t} + \nabla \cdot (\rho_a \mathbf{u}) = 0 \quad (1)$$

for air, and based on the liquid flux in and out of the elemental slice can be given as follows:

$$\frac{\partial \mathbf{c}_v}{\partial t} + \nabla \cdot (\mathbf{u} \mathbf{c}_v - D_{va} \nabla \mathbf{c}_v) = 0 \quad (2)$$

for vapor where the velocity and vapor concentration are known at the inlet ($u_{in} = 1$ m/s, c_v (RH = 50%)) and other boundaries are Neuman zero flux condition. Based on Fick's first law, water vapor flow rate into the elemental volume due to diffusion (mol/s) at any point within the fiber matrix based on the total volume. The porous media here is composed of three volumes of solid matrix (V_s), internal liquid water (V_l), internal vapor water (V_v), and air (V_a). The porosity is defined as all volume component except the solid phase component (i.e., including the liquid phase component and the gas phase component) ratio to all volume components [14].

$$\phi = 1 - \frac{V_s}{V} \quad (3)$$

where the total volume is $V = V_s + V_v + V_l + V_a$. Since the transient internal continuity equations of porous media are

$$\frac{\partial (\phi \rho_a)}{\partial t} + \nabla \cdot (\rho_a \mathbf{u}) = \dot{m}_{evap} \quad (4)$$

based on the superficial velocity of gas for air,

$$\frac{\partial \mathbf{c}_v}{\partial t} + \nabla \cdot (\mathbf{u} \mathbf{c}_v - D_{va} \nabla \mathbf{c}_v) = \dot{m}_{evap} \quad (5)$$

for vapor (negative water vapor mass flow rate),

$$\frac{\partial \mathbf{c}_l}{\partial t} + \nabla \cdot (\mathbf{u} \mathbf{c}_l - D_w \nabla \mathbf{c}_l) = -\dot{m}_{evap} \quad (6)$$

for liquid water where the vapor diffusion coefficient ($D_{va} = 2.6 \times 10^{-5} S_a^{3-\phi} \phi^{2-\phi}$) from the elemental volume due to diffusion, air diffusivity ($D_a = D_{va} (S_a \phi)^{\frac{4}{3}}$), and capillary diffusion coefficient ($D_w = 1.0 \times 10^{-8} \exp(-2.8 + 2c_w/(1-\phi)\rho_s)$) are for porous region. In some references the solved for dry basis moisture content $\left(\frac{(c_l + c_v) M_l}{(1-\phi) \rho_s} \right)$ leads to the same results. The accumulation of water vapor within the gas phase in the elemental slice is also neglected as vapor mass tends to be very low compared to liquid mass at any time during drying. As well the evaporation source phase at the time t is defined proportional to the vapor concentration at porous medium ($c_v = S_v \phi \rho_l / M_l$) and saturated vapor concentration ($c_v = 610.7 \exp[7.5(T_a - 273.1)/(T_a - 35.8)] / RT_a$) as a ratio of volume fractions of air in the pore volume (air saturation, i.e., $S_a = \frac{M_a (1 - \emptyset) \rho_s}{(1 - M_a) \emptyset \rho_s}$) as

$$R_{evap} = K_{evap} \frac{M_l}{RT_a} (c_{v,sat} - c_v) S_a \phi \quad (7)$$

where the air temperature, the gas constant, and the density of the fluid, play a role in that definition along with the evaporation rate constant (K_{evap}). The same definition is valid for porous regions. It is good to notice that vapor pressure and saturated vapor pressure are different and their ratio known as water activity ($\ln \frac{P_v}{p_{sat}(T)} = -0.0267 M_d^{-1.656} + 0.0107 e^{-1.287 M_d} M_d^{1.513} \ln p_{sat}(T)$) is a function of temperature and moisture content on dry basis ($M_d = c_w / (1 - \phi) \rho_s$). Assuming the pressure-driven bulk flow of gas can be given using the Darcy permeability relationship, the air flux in and out can be given as follows:

The transient momentum equation in the free stream is found by applying differential equations for the elemental slice, and considering the accumulation of water in the elemental volume, the continuity equation or mass balance on total moisture (liquid water + water vapor) for the elemental slice can be given as

$$\rho_a \frac{\partial \mathbf{u}}{\partial t} + \rho_a (\mathbf{u} \cdot \nabla) \mathbf{u} = -\nabla p + \nabla \cdot [\mu_a (\nabla \mathbf{u} + (\nabla \mathbf{u})^T) - \frac{2\mu_a}{3} \nabla \cdot \mathbf{u}] \quad (8)$$

pressure-driven bulk flow is due to pressure difference in the gas phase between elemental volumes and the air dynamic viscosity is included. The porous zone transient momentum is defined as

$$\begin{aligned} \frac{\rho_a}{\phi S_a} \frac{\partial \mathbf{u}}{\partial t} + \frac{\rho_a}{\phi S_a} (\mathbf{u} \cdot \nabla) \mathbf{u} \frac{1}{\phi S_a} = -\nabla p + \nabla \cdot \left[\frac{\mu_a}{\phi S_a} (\nabla \mathbf{u} + (\nabla \mathbf{u})^T) \right. \\ \left. - \frac{2}{3} \frac{\mu_a}{\phi S_a} (\nabla \cdot \mathbf{u}) \right] - \frac{C_F}{k^{1/2}} \rho |u| u - (\mu_a/k_{ar} + \mu_w/k_{wr}) \mathbf{u} \end{aligned} \quad (9)$$

the pressure-driven bulk flow of gas inside porous can be given using the Darcy permeability relationship. The pressure in the elemental volume is a moisture content or saturation dependent on the elemental volume. Assuming the Darcy model with permeability correlation for each phase is defined as a function of volume fractions of water in the pore volume (water saturation, i.e., $S_w = \frac{M_w (1 - \phi) \rho_s}{(1 - M_w) \phi \rho_s}$ and $S_a + S_w = 1$) as [49,50]

$$k_{ar} = \begin{cases} 1 - 1.1 S_w, & S_w < 1/1.1 \\ 0, & S_w > 1/1.1 \end{cases} \quad (10)$$

which is for gas flow is related to the saturated permeability when the porous media is fully wet and a relative permeability k_{ar} and was obtained from experiments.

$$k_{wr} = \begin{cases} \left(\frac{S_w - 0.08}{0.92} \right)^3, & S_w > 0.08 \\ 0, & S_w < 0.08 \end{cases} \quad (11)$$

where the density of the solid matrix, water, and air molecular weight plays a role in that definition.

The energy balance in an elemental volume can be derived using Fourier's law of heat conduction and convective heat transfer due to the vapor and liquid movement into and out of the elemental volume as follows. The heat transfer equation in the free stream is defined as

$$\frac{\partial (\rho_a c_{p,a} T_a)}{\partial t} + \nabla \cdot (\rho_a c_{p,a} \mathbf{u} T_a) = \nabla \cdot (k_a \nabla T_a) \quad (12)$$

The energy for condensation in an elemental volume is given by the net accumulation of water vapor in an elemental slice due to vapor diffusion and bulk flow. This assumes that there is no accumulation of water vapor within the gas phase in an elemental volume and if there is any net accumulation it is assumed to be condensed into liquid. The heat transfer equation in the porous zone based on effective thermophysical parameters (calculated by weighting each component) and source term with latent heat of evaporation is defined as

$$\begin{aligned} \frac{\partial ((\rho_s c_{p,s} (1 - \phi) + \rho_l c_{p,l} \phi S_l + \rho_a c_{p,a} \phi (1 - S_l)) T_m)}{\partial t} + \nabla \cdot ((\mathbf{u} \rho_s (1 - \phi)) c_{p,s} T_m + (\mathbf{u} c_l - D_{cap} \nabla c_l) c_{p,l} T_m \\ + (\mathbf{u} \rho_a + \mathbf{u} c_a) c_{p,a} T_m) = \nabla \cdot (((1 - \phi) k_s + S_l \phi k_l + \phi k_a (1 - S_l)) \nabla T_m) - R_{evap} h_{evap} \end{aligned} \quad (13)$$

where the temperature is known at the inlet ($T_{in} = 80^\circ\text{C}$) and other boundaries are Neuman zero flux condition. In the above equation, the liquid and vapor phases are assumed to be in thermodynamic equilibrium and hence the vapor pressure in the gas phase is given by the water-vapor saturation pressure relationship. [Table 1](#) shows some parameters used in the simulations while [Table 2](#) shows thermodynamically constants of the system.

Table 1: Coefficients used in governing equations

Parameter	Value
c_{ps} (J/kg K)	1566
c_{pw} (J/kg K)	4180
D_w (m ² /s)	$1.0 \times 10^{-8} \exp(-2.8 + 2.0 M)$
$D_{\text{eff},g}$ (m ² /s)	$2.6 \times 10^{-5} \varepsilon^{4/3} S_g^{10/3}$
h (W/m ² K)	20
h_{mv} (m/s)	0.01
k_s (W/m K)	0.21
k_w (W/m K)	0.64
k_g (W/m K)	0.026
S_{wi}	0.5
S_{ir}	0.09
T_{amb} (°C)	20
T_i (°C)	20
ϕ	0.75
λ (J/kg)	2.435×10^6
μ_g (Pa s)	1.8×10^{-5}
μ_w (Pa s)	5.468×10^{-4}
ρ_s (kg/m ³)	1419
ρ_w (kg/m ³)	1000

Table 2: Thermodynamic properties

Parameter	Value	Unit
Water		
Conductivity of water	0.680168	W/m/K
Density of water	960	kg/m ³
Heat of formation	1.5866E7 star	J/kg
Molecular weight of water	18.0153	kg/kmol
Dynamic viscosity of water (temperature)–field function	8.89E–04	Pa s
Saturation pressure of water (25°C)	Antoine equation	Pa
Saturation temperature of water (real temperature)	373.12	K
Standard state temperature	298.15	K

(Continued)

Table 2 (continued)

Parameter	Value	Unit
Critical temperature of the water	647.12	K
Ideal gas constant	8.314	J/mol/K
Air		
Density of air	Ideal gas	kg/m ³
Conductivity of air	0.031672	W/m/K
Specific heat capacities of air	991.608	J/kg/K
Molecular weight of air	0.0289664	kg/mol
Dynamic viscosity of air	1.86E-05	Pa s
Standard state temperature	298.15	K
Vapor		
Vapor thermal conductivity	0.027518	W/m/K
Vapor specific heat capacities	1966.48	J/kg/K
Heat of formation	1.34234E7	J/kg
Molecular weight	0.0181053	Kg/mol
Standard state temperature	298.15	K
Dynamic viscosity of vapor (temperature)	1.27E-05	Pa s
Potato		
Potato conductivity	0.138457	W/m/K
Potato specific heat capacities	1338.88	J/kg/K
Potato density	1200	kg/m ³
Porosity of the potato	0.72	–

2.2 Analytical Solutions

Generally, in addition to the above heat and mass transfer equations, it is also important to consider the momentum balance equations for liquid and gas to solve for the velocity of the two phases. This can be considered using the Navier-Stokes equation including the viscous and inertial terms and the pressure difference and gravity. In paper drying applications, in general, at low fluid velocities, the inertial terms can be considered to be negligible. Hence, neglecting gravity, pressure terms, and viscous terms in momentum cannot make a big difference. This essentially reduces to Darcy's for flow through porous media described earlier in equations. It should be noted again here that the liquid and gas phase permeability and capillary pressure are functions of moisture content or saturation and should be known from experimental results. The Darcy's equation is

$$0 = -\nabla P - \frac{\mu}{k}u + \frac{\mu_c}{\phi} \frac{\partial^2 u}{\partial z^2} \quad (14)$$

Darcy's law was chosen for implementation because it is a well-established and widely used equation in fluid dynamics. It provides a simplified and practical approach to model the flow of fluids through porous media, such as the subsurface of Venus. By using Darcy's law, researchers can simulate and analyze the movement of fluids, including the transport of heat and mass, within Venus' atmosphere. Its application allows for a better understanding of the dynamics and behavior of the atmosphere, aiding in the study of various phenomena, such as the circulation patterns and

the influence of solar radiation. In addition to the liquid and vapor phases of water, we also need to consider the mass balance for air within and outside the material. The basic assumption here, as stated earlier, is that there is counter diffusion of air and pressure-driven bulk flow of air (along with water vapor in the gas phase). Likewise, we consider local thermal equilibrium and all the solid, liquid, and gas phases are at the same temperature; the liquid water and water vapor are in thermodynamic equilibrium. Based on these assumptions, we can write the following equations. The velocity continuity at the porous interface implies that the analytical solution outside the porous media ($z \in [d, d + H]$) as

$$u = - \left(\frac{H^2}{4\mu} \right) \nabla P \left[- \frac{(z-d)^2}{H^2} + (2-U) \frac{z-d}{H} + U \right] \quad (15)$$

where

$$U = \frac{1 - \operatorname{sech} \chi \delta + \alpha \tanh \chi d}{\beta N \chi^2} \quad (16)$$

$$\chi^2 = \frac{L^2}{MK} \quad (17)$$

$$N = \frac{\mu_e}{\mu} \quad (18)$$

$$\beta = \left(1 + \frac{\tanh \chi d}{2N\chi} \right) \quad (19)$$

Drying can be defined as removing a solvent through evaporative mass transfer. Evaporation involves a liquid and a gas phase, so it is essentially a multiphase problem. Since evaporation can occur inside the solid (treated as porous media), the easiest approach to model this problem is using the mixture multiphase model. In order to conduct a thorough numerical simulation, first a set of simplified governing equations describing the mass conservation, momentum conservation, and energy conservation for the liquid and gas (comprising water and air). This solution later will be used to extract the Nusselt number at the boundary of the porous medium. Similar to the liquid continuity equation where we considered the water vapor flux due to diffusion ($\mathbf{J}_X = -D_m \nabla X - \delta D_m \nabla T$), here we will consider the flow due to diffusion and convection in an elemental volume. The mass transfer balance

$$\frac{\partial X}{\partial t} = D \frac{\partial^2 X}{\partial z^2} + D\delta \frac{\partial^2 T}{\partial z^2} \quad (20)$$

The thermal balance (with a heat source of a mass transfer)

$$\frac{\partial T}{\partial t} = \alpha \frac{\partial^2 T}{\partial z^2} + \frac{\epsilon h_{\text{evap}}}{C_p} \frac{\partial X}{\partial t} \quad (21)$$

with symmetry boundary condition at $z = 0$

$$\frac{\partial T}{\partial z} = \frac{\partial X}{\partial z} = 0 \quad (22)$$

and flux boundary condition at $z = d$

$$-\lambda \frac{\partial T}{\partial z} = h(T - T_a) \quad (23)$$

It can also be assumed there is no liquid accumulation at the open surface boundary and whatever liquid is able to move towards the open surface at the surface layer is being pushed out of the sheet.

$$-D \frac{\partial X}{\partial z} = h_m(X - X_a) + D\delta \frac{\partial T}{\partial z} \quad (24)$$

which is non-dimensionalized to the

$$\frac{\partial X^*}{\partial Fo} = Lu \frac{\partial^2 X^*}{\partial Z^2} + LuPn \frac{\partial^2 \theta}{\partial Z^2} \quad (25)$$

$$\frac{\partial \theta}{\partial Fo} = \frac{\partial^2 \theta}{\partial Z^2} + \varepsilon Ko \frac{\partial X^*}{\partial Fo} \quad (26)$$

with a non-dimensionalized symmetry boundary condition at $Z = 0$

$$\frac{\partial \theta}{\partial Z} = \frac{\partial X^*}{\partial Z} = 0 \quad (27)$$

RH in the equation is the relative humidity at a given moisture content of the sheet. This gives the percent relative humidity of the surrounding atmosphere at which paper at a certain moisture content will be in equilibrium. The vapor pressure reduction is also related to the capillary pressure in unsaturated porous media and for a porous media with wettable liquid (concave meniscus). For specific types of porous media, the vapor pressure reduction relationships are also obtained from adsorption isotherms of water vapor in cellulosic paper and board materials. Then the boundary condition non-dimensionalized flux boundary condition at $Z = 1$.

$$\frac{\partial X^*}{\partial Z} = Bi_m(X_{a,in}^* - X) - Pn \frac{\partial \theta}{\partial Z} \quad (28)$$

$$\frac{\partial \theta}{\partial Z} = Bi_t(\theta_{a,in} - \theta) \quad (29)$$

By use of the spatial average and zero-order Hermite's approximation analytical solution is obtained as follows:

$$X = e^{\lambda_1 Fo} \gamma_1 + e^{\lambda_2 Fo} \gamma_2 + \gamma_3 \quad (30)$$

$$\theta = \lambda_1 \kappa_1 (1 + \beta_1) e^{\lambda_1 Fo} + \lambda_2 \kappa_2 (1 + \beta_1) e^{\lambda_2 Fo} + \kappa_3 \quad (31)$$

where

$$\lambda_1 = \frac{\alpha_1 + \beta_2 + \sqrt{\Delta}}{2} \quad (32)$$

$$\lambda_2 = \frac{\alpha_1 + \beta_2 - \sqrt{\Delta}}{2} \quad (33)$$

$$\gamma_1 = \frac{-\frac{1}{4}(-\alpha_1 \sqrt{\Delta} - \beta_2 \sqrt{\Delta} + \Delta)(-\alpha_1 \beta_2 - \alpha_1 \beta_3 + \beta_2^2 + \beta_2 \beta_3 + \beta_2 \sqrt{\Delta} + \sqrt{\Delta} \beta_3 + 2\alpha_3 \beta_1 + 2\alpha_2 \beta_1)}{((-\beta_1 \alpha_2 + \alpha_1 \beta_2) \Delta)} \quad (34)$$

$$\gamma_2 = \frac{-\frac{1}{4}(\alpha_1 \sqrt{\Delta} + \beta_2 \sqrt{\Delta} + \Delta)(-\alpha_1 \beta_2 - \alpha_1 \beta_3 + \beta_2^2 - \beta_2 \sqrt{\Delta} + \beta_2 \beta_3 - \sqrt{\Delta} \beta_3 + 2\beta_1 \alpha_2 + 2\beta_1 \alpha_3)}{((-\beta_1 \alpha_2 + \alpha_1 \beta_2) \Delta)} \quad (35)$$

$$\gamma_3 = \frac{\beta_1 \alpha_3 - \alpha_1 \beta_3}{-\beta_1 \alpha_3 + \alpha_1 \beta_2} \quad (36)$$

$$\kappa_1 = \frac{-\frac{1}{4} \left(\frac{1}{2} \beta_2 + \frac{1}{2} \alpha_1 + \frac{1}{2} \sqrt{\Delta} \right) (-\alpha_1 \sqrt{\Delta} - \beta_2 \sqrt{\Delta} + \Delta) (-\alpha_1 \beta_2 - \alpha_1 \beta_3 + \beta_2^2 + \beta_2 \beta_3 + \beta_2 \sqrt{\Delta} + \sqrt{\Delta} \beta_3 + 2\alpha_3 \beta_1 + 2\alpha_2 \beta_1)}{\beta_1 ((-\beta_1 \alpha_2 + \alpha_1 \beta_2) \Delta)} \quad (37)$$

$$\kappa_2 = \frac{-\frac{1}{4} \left(\frac{1}{2} \beta_2 + \frac{1}{2} \alpha_1 + \frac{1}{2} \sqrt{\Delta} \right) (\alpha_1 \sqrt{\Delta} + \beta_2 \sqrt{\Delta} + \Delta) (-\alpha_1 \beta_2 - \alpha_1 \beta_3 + \beta_2^2 + \beta_2 \beta_3 - \beta_2 \sqrt{\Delta} - \sqrt{\Delta} \beta_3 + 2\alpha_3 \beta_1 + 2\alpha_2 \beta_1)}{\beta_1 ((-\beta_1 \alpha_2 + \alpha_1 \beta_2) \Delta)} \quad (38)$$

$$\kappa_3 = \beta_2 \left(\frac{-\beta_3 \alpha_1 + \alpha_3 \beta_1}{-\alpha_2 \beta_1 + \alpha_1 \beta_2} \right) + \beta_3 \quad (39)$$

$$\Delta = (\alpha_1 - \beta_2)^2 + 4\alpha_2 \beta_1 \quad (40)$$

$$\alpha_1 = -(1 + 2\varepsilon KoLuPn) \frac{4}{Bi_t + 4} - \frac{4Bi_m LuPnBi_t}{(Bi_m + 4)(Bi_t + 4)} \quad (41)$$

$$\alpha_2 = -\frac{4\varepsilon KoLuBi_m}{Bi_m + 4} \quad (42)$$

$$\alpha_3 = -\alpha_1 \theta_a - \alpha_2 X_a^* \quad (43)$$

$$\beta_1 = \frac{-4PnBi_t LuBi_m}{(Bi_m + 4)(Bi_t + 4)} \quad (44)$$

$$\beta_2 = \frac{-4LuBi_m}{Bi_m + 4} \quad (45)$$

$$\beta_3 = -\beta_1 \theta_0 - \beta_2 X_0^* \quad (46)$$

2.3 Multi-Agent Simulation

In this part, multi-agent reinforcement learning is introduced and the other methods used (genetic algorithms, pattern search, and particle swarm optimization) are briefly explained. Genetic algorithms (GA) are a type of optimization algorithm inspired by the process of natural selection. They use a population of potential solutions and apply selection, crossover, and mutation operations to evolve and improve the solutions over generations. GAs are often used to solve complex optimization problems where traditional methods may struggle. On the other hand, multiagent reinforcement learning is a field of machine learning that focuses on training multiple agents to interact and learn from their environment. Each agent learns through trial and error, using reinforcement learning techniques such as rewards and punishments. MARL is used to study how multiple agents can collaborate or compete to achieve specific goals. Both genetic algorithms and multiagent reinforcement learning have their strengths and applications. GAs are great for optimization problems, while MARL is useful for studying complex interactions and coordination among multiple agents.

Pattern search (PS) is a local optimization method that explores the search space by iteratively moving from one point to another, searching for the best solution. It can be efficient when the search

space is small or when the objective function is relatively simple. However, it may struggle with complex or high-dimensional problems, as it can get stuck in local optima. On the other hand, MARL involves multiple agents learning and interacting with their environment simultaneously. The speed of finding a solution in MARL depends on various factors, such as the complexity of the environment, the number of agents, and the learning algorithms used. In some cases, MARL can lead to faster solutions compared to pattern search, especially when dealing with complex problems that require coordination and collaboration among agents. In summary, pattern search can be faster for simple optimization problems with small search spaces, while MARL has the potential to find faster solutions for complex problems that involve multiple agents and interactions.

Particle swarm optimization (PSO) is a population-based optimization algorithm inspired by the behavior of bird flocking or fish schooling. It uses a group of particles that move through the search space, adjusting their positions based on their own best-known position and the global best-known position. PSO can be efficient in finding solutions for optimization problems, especially when the search space is large or complex. On the other hand, MARL involves multiple agents learning and interacting with their environment simultaneously. The speed of finding a solution in MARL depends on various factors, such as the complexity of the environment, the number of agents, and the learning algorithms used. In some cases, MARL can lead to faster solutions compared to PSO, especially when dealing with complex problems that require coordination and collaboration among agents. In summary, particle swarm optimization can be fast in finding solutions for optimization problems, while MARL has the potential to find faster solutions for complex problems that involve multiple agents and interactions.

As noticed in the introduction section, MARL is not trendy in the field. Therefore, a straightforward model environment (e.g., grid environment with 0.01 for the first examination is considered) is used for the study to conduct comparative experiments. In the introduction, section, a literature search of various MARL methods of the real-world situation was introduced. On the other hand, there are necessities for an intuitive interface and low computational cost optimization for communicating the environment-model and which are significant to attain the efficient method (see Fig. 2). In general, the flowchart of a MARL (Multi-Agent Reinforcement Learning) algorithm typically consists of several steps. These steps may include:

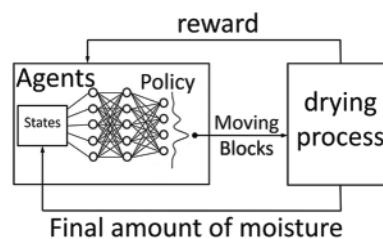


Figure 2: Flowchart of MARL used here

1. Initialization: Initializing the environment, agents, and any necessary parameters.
2. Observation: Agents observe the current state of the environment and gather relevant information.
3. Action Selection: Agents select actions based on their observations and a policy, which could be a learned policy or a predefined one.
4. Interaction: Agents interact with the environment by executing their selected actions.

5. **Reward Calculation:** Agents receive feedback in the form of rewards or penalties based on their actions and the state of the environment.

6. **Learning:** Agents update their policies based on the observed rewards and the chosen learning algorithm, such as Q-learning or Deep Q-Networks (DQN).

7. **Iteration:** The process of observation, action selection, interaction, reward calculation, and learning continues for multiple iterations or episodes.

The flowchart of MARL may vary depending on the specific algorithm or approach being used. It is important to refer to the specific details mentioned in the manuscript to get a more accurate understanding of the flowchart and the function of each step.

3 Results

The manuscript does not explicitly mention the methodology used for computational modeling. It would be helpful to have more information or clarification from the authors regarding the specific technique employed, such as whether it utilizes the Finite Element Method or another computational approach. Understanding the methodology used is crucial for evaluating the accuracy and reliability of the computational modeling results. In this section the finite element computational tool established in a homemade code is used to solve laminar flow with mass transfer was used. [Table 2](#) presents the geometric parameters of the problem. A convergence criterion of 10^{-6} is selected.

3.1 Analytical Simulation

In this section, the results of the previous section are discussed and explained. For simplicity, a 2D approximation of a sheet of paper inside a tunnel is chosen as a solution domain. Dry air enters the inlet, evaporates the water present in the porous solid region, and exits through the outlet. The porous solid region is specified at $1 - \phi$ volume fraction of water as an initial condition. The geometry of the problem is plotted in [Fig. 1](#) with the parameters of [Table 1](#). [Table 2](#) presents the properties of water, air, and vapor respectively which are used in the current study. The preliminary data shown here indicate that the drying of porous media involves a complex balance between local heat transfer, vaporization and condensation, and liquid and vapor fluxes. This is also very much a function of the boundary conditions and the dried porous material properties.

[Fig. 3](#) presents velocity profiles for various porous thicknesses in the tunnel. As shown thicker porous media inside the channel can change the velocity profile inside it. It can be noticed that the porous layer compresses the maximum velocity location towards the free part of the channel from the centerline. In addition, the values of velocity at the interface (relative to the maximum velocity). Besides, [Fig. 3](#) shows the critical role of the height of the non-porous medium in the velocity profile. The above figure can be used for the calculation of boundary conditions for heat and mass transfer of any analytical solution inside the porous medium.

An input relative humidity of 0% might not accurately reflect real-world conditions. Adjusting the assumed relative humidity could potentially bring the simulation results closer to the actual results. It is always important to consider environmental factors when interpreting simulation data. The convective drying of porous media modeled by the current study provides reasonable results. As revealed in [Fig. 4](#) The number of the system increased as the velocity ratio over the porous layer increased. The velocity ratio is the velocity at the interface relative to the maximum velocity in the channel. Given the initial higher temperature of the sheet (373 K), there is no warm-up period observed. Based on those properties Schmidt Number (Molecular diffusivity) is 0.6, and Turbulent Prandtl is neglected.

Given the hot surface boundary on one side there is greater conductive heat flux closer to the hot surface and then decreases along the porous layer thickness. This essentially contributes to the overall drying rate kinetics observed. The Reynolds analogy is the common model to estimate the Prandtl number when performing CFD simulations. It expresses a similarity between momentum exchange and heat transfer in a fluid. The Nusselt number, as a dimensionless word of the warmth transfer coefficient ($Nu = \partial(T/\Delta T)_w \partial(n/C)_w$) and the Sherwood number, for the mass transfer coefficient ($Sh = \partial(c/\Delta c)_w \partial(n/C)_w$), are then functioning of those limits. They are the dimensionless heat or mass concentration gradients at the porous surface.

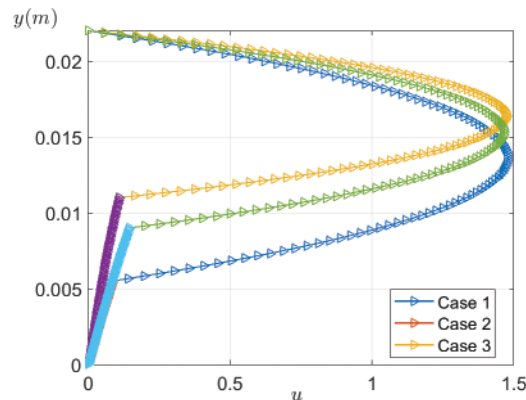


Figure 3: Velocity profiles for various porous thicknesses in the tunnel

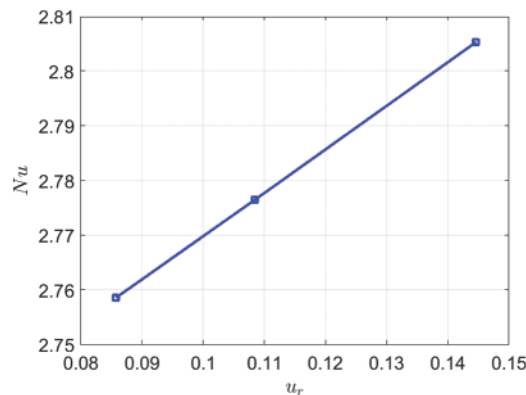


Figure 4: Nusselt number of the system as a function of velocity ratio over porous layer

It is good to mention that Fig. 1a has the same labeling of the length and height of the porous media as it is used for a benchmark of experimental setup where a slice of sample with known weight and height is used. The initial condition of the dry air in the system is $RH = 0.5$. Table 3 presents the geometrical parameters of the benchmark problem.

Fig. 5 provides a comparison of analytical, CFD, and experimental results. The thin layer of potato is also can be used for de-icing and defogging simulations using fluid film thin film best practices case. Results of the transient problem converged at a time fraction of 0.8 s are plotted in Fig. 5. The moisture is approximated using a forward difference technique. This shows a more gradual warming up of the porous media with the resultant constant drying rate period and a falling drying rate period. As moisture based on the solid mass is the mass of liquid water per mass of solid and the mass of liquid

water equals the density of liquid water times volume of drying region times porosity times volume fraction of liquid water (for the solid 1-porosity should be used) since the initial value of the moisture based on the solid mass could be higher than 1. The drying rate is also can be derived from the Fig. 5. It equals the mass loss of liquid water per drying time times surface area or moisture gradient times the mass of solid divided by surface area. The comparison shows that the current study predicts higher values of drying rate than the experimental. The plots confirm the presence of liquid flux from the hot surface side to the open surface side and the magnitude of the flux decreases as drying proceeds. After the constant rate period, the internal temperature porous medium begins to increase indicating that some of the energy transferred is being used for sensible heating. This continues until the drying rate falls to almost zero and the porous medium temperature reaches close to the air temperature.

Table 3: Geometrical parameters of the benchmark problem

Symbol	Description	Value	Unit
W_p	Potato thickness	0.01	m
L_p	Potato length	0.1	m
W	Tunnel height	0.15	m
L	Tunnel length	0.5	m

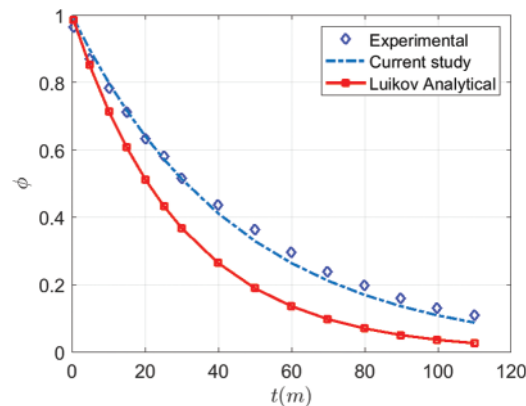


Figure 5: Comparison of analytical, CFD, and experimental results

3.2 Multi-Agent Simulation

It would be helpful to have a comparison of the calculation accuracy of various algorithms in addition to the calculation speed. It would provide a more comprehensive understanding of the advantages of the MARL algorithm in a large space. Including accuracy in the comparison would give a clearer picture of how well the MARL algorithm performs compared to other algorithms. Comparing the simulation results of MARL with other algorithms for the optimal position would indeed highlight the advantages of MARL. It would provide a direct comparison to showcase how well MARL performs in finding the best position compared to other algorithms. Including this comparison would strengthen the argument for the effectiveness of MARL in optimizing positions. In Fig. 1b where different values are depicted. By considering that, we can see if it has any significant impact on the results. It is important to have accurate and consistent labeling to ensure clarity in the figures. The

initial condition of the dry air in the system is $RH = 0.5$ and all properties same as the inlet condition while the time step is automatically controlled by software. As well the porous viscous resistance of water and gas is considered as 10^9 . Usually, the wall treatment models (blended wall function) are used to set near-wall modeling assumptions for turbulence models but here neglected. The static temperature considered here condition (see Table 1) is the temperature of the gas if it had no ordered motion and was not flowing. The geometric parameters of the optimization problem are mentioned in Table 4. Usually, a simple grid division with 0.01 for the first examination is considered to conduct comparative experiments. Table 5 shows the grid independence test used in current computational fluid dynamics analyses to ensure that the results obtained are not dependent on the grid used for the simulation. Grid dependent solutions shown here lead to a potentially costly engineering design decisions.

Table 4: Geometrical parameters of the optimization problem

Symbol	Description	Value	Unit
a	Potato thickness	0.02	m
a	Potato length	0.02	m
W	Tunnel height	0.05	m
L	Tunnel length	0.7	m

Table 5: Grid study check

Grid	Average relative error
$\Delta x = \Delta y = 0.1$	3.46×10^{-3}
$\Delta x = \Delta y = 0.1$	3.46×10^{-3}
$\Delta x = \Delta y = 0.1$	3.46×10^{-3}
$\Delta x = \Delta y = 0.1$	3.46×10^{-3}

PSO and GA are population-based methods that explore the search space efficiently, making them suitable for large-scale optimization problems. Pattern search, although a local optimization method, can also be effective for large problems if the search space is well-defined. MARL, on the other hand, involves multiple agents learning and interacting with their environment. The complexity of coordinating multiple agents and the increased computational requirements can make it slower compared to PSO, pattern search, and GA for large-scale problems. In summary, for large-scale problems, PSO, pattern search, and GA are generally faster than MARL due to their specific optimization approaches and computational requirements.

In terms of memory, PSO and pattern search typically require less memory compared to GA and MARL. PSO and pattern search methods usually store the positions and velocities of particles or search patterns, respectively. These methods do not require extensive memory for storing large populations or complex agent interactions. On the other hand, GA and MARL may require more memory due to their population-based or multi-agent nature. GA involves maintaining a population of individuals, which can consume more memory as the population size increases. MARL requires memory to store the state and policies of multiple agents, as well as their interactions and observations. In summary, PSO and pattern search methods generally require less memory compared to GA and

MARL, which may require more memory due to their population-based or multi-agent approaches. When it comes to accuracy, let us compare the methods of particle swarm optimization (PSO), pattern search, genetic algorithms (GA), and multi-agent reinforcement learning (MARL).

In terms of accuracy, it is important to note that the performance and accuracy of these methods can vary depending on the specific problem and implementation. However, in general:

- PSO and pattern search methods are deterministic and can provide accurate results if the search space is well-defined and the objective function is smooth and continuous.

- GA is a probabilistic method that uses evolutionary principles and can provide reasonably accurate solutions, especially for optimization problems with a large search space and multiple global optima.

- MARL, being a learning-based approach, can achieve high accuracy over time as agents interact with the environment and learn optimal policies. However, achieving high accuracy in MARL can be challenging and may require a longer learning process.

In summary, PSO and pattern search methods can provide accurate results for well-defined problems, GA can achieve reasonably accurate solutions for large search spaces, and MARL can achieve high accuracy over time through learning and interaction with the environment.

The warmth and mass transfer equations for the porous moist objects are coupled with the channel movement equations by utilizing the favorable spacing to accomplish the best convective drying efficiency. The moisture reduction of the objects is higher at the optimum spacing in comparison with unsteady replication outcomes acquired from resolving the parametric variation of the unsteady coupled discipline equations. Fig. 6 presents the essays on each method used to reach the optimal point of the system. As shown in Fig. 6, the comparison of the results of various algorithms for optimizing the convective drying proves that MARL is the fastest method.

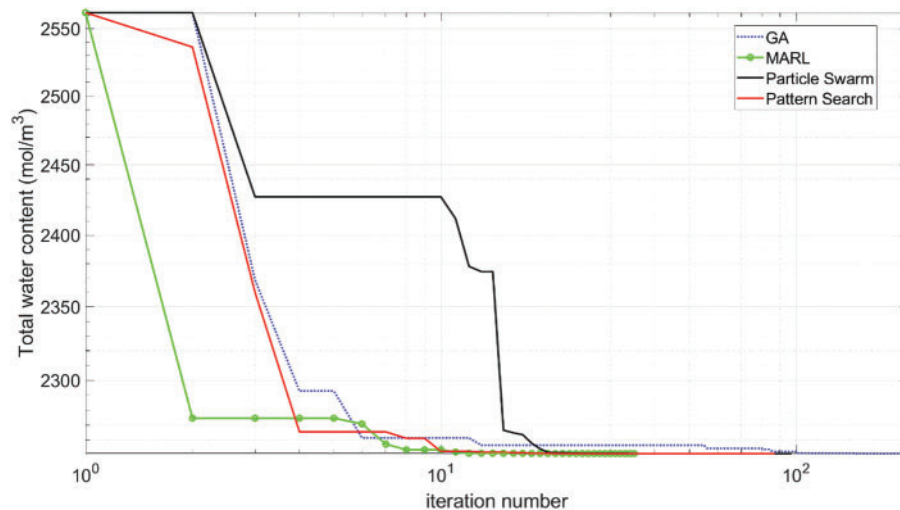


Figure 6: Comparison of the results of various algorithms on optimizing the convective drying

Even though, initially, the blocks have a uniform liquid saturation, as drying proceeds a non-uniform saturation profile develops within the blocks. This is owing to the temperature and pressure profiles within the sheet and the corresponding vaporization/condensation and liquid and vapor fluxes within the blocks. There is pressure developed within the blocks greater than the atmospheric pressure

and the pressure is higher closer to the hot surface boundary and decreases towards the open surface (Fig. 6a). The temperature profile within the blocks follows a similar profile from the hot surface side to the open surface side (Fig. 7b). Fig. 7a shows the pressure contours inside the fluid and porous system. The existence of the porous domain causes the highest pressure to change in the domain. Fig. 7b shows the velocity contours in the system as a laminar confined flow. Inside the porous domain, the velocity is close to the wall (see Fig. 1b). Fig. 7c shows the temperature change in the system as a significant cooling in the whole domain. Fig. 7d illustrates the concentration (a measure of relative humidity) after 10000 s. As evaporation takes place internal to the porous media it experiences lower temperatures. The relative humidity in the porous medium goes from almost 100% at the beginning of the simulation to ambient humidity by the end. As revealed the dry air pushes the moist air out of the potato. Since moisture in the porous medium is a mix of liquid and vapor water, a measure of the drying progression is liquid saturation. As the water evaporates, it lowers the temperature of the porous medium. As shown taller porous blocks expand the moisture through the domain and have enhanced dealings with surroundings. As exposed by the augment of the distance less moisture is received at the preceding block. Additionally, the smaller distances assist the potatoes to maintain their wet content which is not most wanted in this purpose. As well the first block is affected too much by the environmental condition. Similarly, because of the moisture gradient along the horizontal (x-axis) and thickness direction (y-axis), there is also liquid and gas flow along the horizontal or x-axis, although the magnitude of these fluxes is comparatively lower.

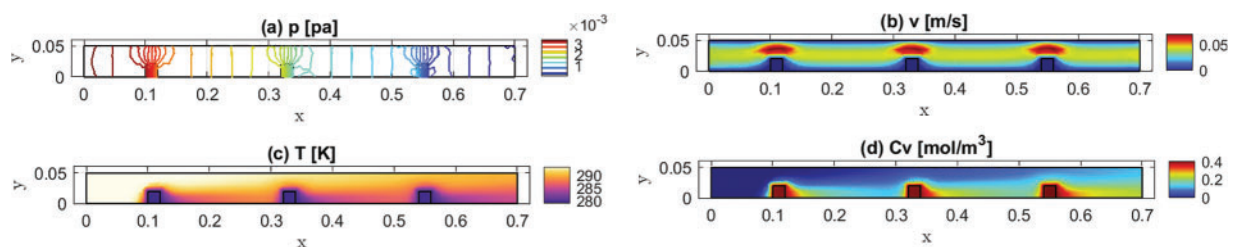


Figure 7: (a) Pressure, (b) velocity magnitude, (c) temperature, (d) vapor concentration after 10000 s for uniform arrangement

Fig. 8a shows the logarithm pressure contours inside the fluid and porous system. The existence of the porous domain at the inlet causes the highest pressure to change in the domain for optimal arrangement. Fig. 8b shows the Velocity contours in the system as a laminar confined flow. Inside the porous domain, the velocity is close to the wall (see Fig. 1b). Fig. 8c shows the temperature change in the system as a significant cooling in the whole domain after 10000 s for optimal arrangement. In comparison with Fig. 7c, the effect of arrangement can affect inlet heat to the utmost temperature variation in the system. As publicized, less inlet temperature is caused by optimal arrangement. Such dependence shows dependence on thermal balance. The internal block temperature also remains constant during the constant drying rate period indicating that all the energy transferred from the flowing air is used for evaporation of the moisture and not for sensible heating of the block. Fig. 8d illustrates the concentration (a measure of relative humidity) after 10000 s. The relative humidity in the porous medium goes from almost 100% at the beginning of the simulation to ambient humidity by the end. As shown the dry air pushes the wet air out of the potato. It is also faster near the surfaces exposed to the ambient air than inside the medium. As given away vapor concentration on the surface is decreased by increasing height. As made known, it maintains at a steady level, and following some elevation it decreases suddenly. Corresponding to the vapor saturation profile, there is

also a non-uniform gas saturation profile through thickness and the horizontal axis. Corresponding to the liquid saturation profile, the next plot can show this effect (see Fig. 9).

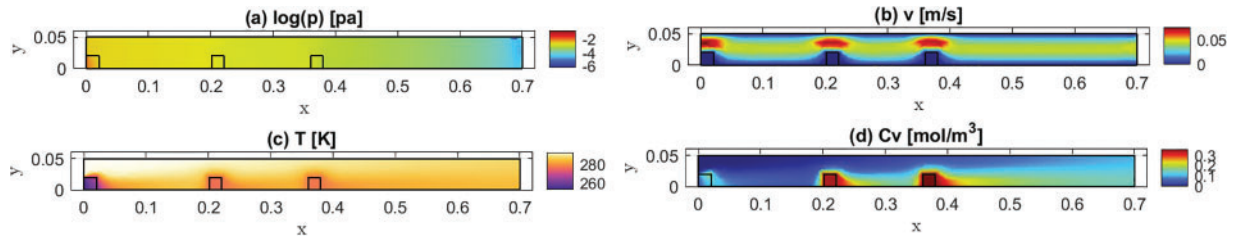


Figure 8: (a) Pressure, (b) velocity magnitude, (c) temperature, (d) vapor concentration after 10000 s for optimal arrangement

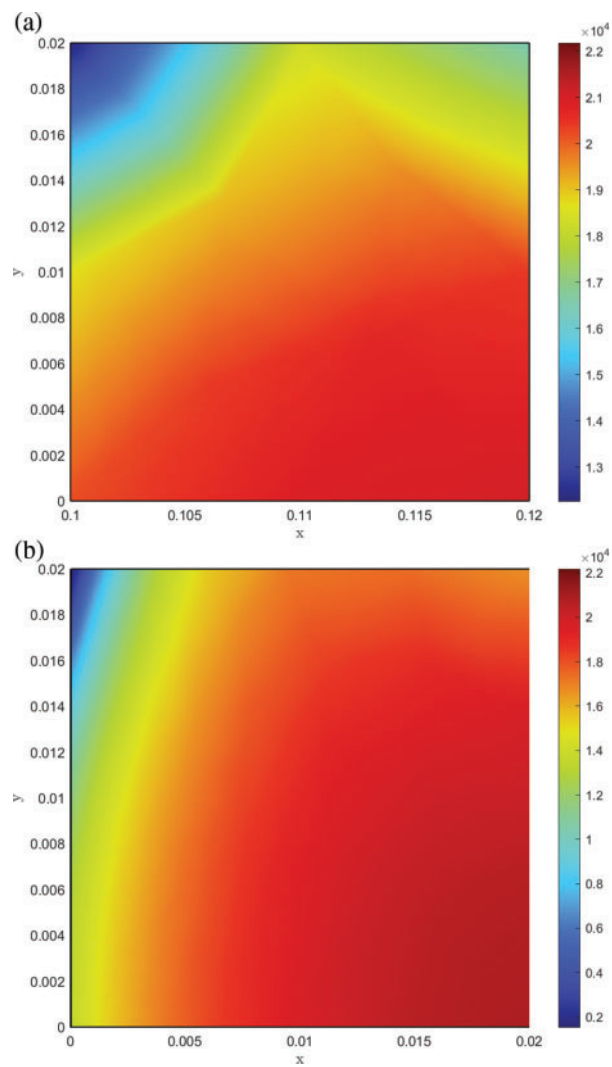


Figure 9: Liquid concentration in the first block after 10000 s for (a) uniform arrangement and (b) optimal arrangement

Velocity, volume fraction of water, evaporation rate, and mass fraction of steam are affected by each other simultaneously. Fig. 9 presents liquid concentration in the first block after 10000 s for uniform and optimal arrangement. The saturation (volume fraction of liquid water) and moisture based on the solid mass can be described as a measure of drying in the system. As shown by the increase in height from the bottom less humidity remained at the block. Liquid water and water vapor evaporated inside the porous media can be transported to the surface/outside of the porous solid region and then convected by the external air stream. Additionally, the left of the first block sooner emptied from the water while uniform arrangement helps the potato to maintain its moist.

4 Conclusions

In this study, a comprehensive model was developed for drying potatoes with the implementation of MARL to optimize slice locations. As well this research focus was on drying problems in a porous media. A multiphase model is designed to simulate the drying of potatoes by convective heat transfer with porous media techniques. To do that a set of governing equations including mass conservation, momentum conservation, and energy conservation for liquid and vapor phase in thermal equilibrium with solid matrix are solved by homemade software. The results are compared by experimental and analytical methods of drying Flow Benchmark. As demonstrated MARL has a better efficiency regarding iteration needed rather than previous methods. As well porous media increase the convective mass and heat transfer through the domain. It is renowned that this mathematical study investigated the viability, as well as the predictable shape of porous blocks, porous distribution inside the system, shrinkage, and variable diffusivity that can affect the results which are recommended for further studies. The main outcome of this research can be summarized as:

- PSO and PS methods (local and deterministic) generally require less memory compared to GA (probabilistic and population-based which uses evolutionary principles) and MARL (agent interactions).
- PSO and PS methods provide accurate results but here the objective function is not smooth even is continuous. Since GA and MARL provide reasonably accurate solutions, as current problem has a large search space and requires learning and interaction with the environment.
- PSO and PS could be faster than GA and MARL based on the initial position near the optimal solution while MARL shows its capability to understand the different flow regimes by the interaction of various agents with the environment.

To do more research here adding the slope of potatoes is recommended. If the porous medium is situated on a surface with a slight slope, it could have an impact on the calculations and modeling. The slope introduces an additional factor that needs to be considered, as it affects the flow dynamics and the distribution of fluids within the porous medium. The presence of the slope can alter the pressure gradients and flow velocities, potentially leading to variations in the transport of heat and mass. To accurately model and calculate the behavior of fluids in such a scenario, the slope of the surface would need to be incorporated into the mathematical equations and simulations. This would allow for a more precise representation of real-world conditions and improve the accuracy of the results obtained from the calculations.

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Author Contributions: Study conception and design; data collection; analysis and interpretation of results; draft manuscript preparation is performed by MYAJ. The author reviewed the results and approved the final version of the manuscript.

Availability of Data and Materials: All data used in the study are available upon request.

Conflicts of Interest: The author declares that he has no conflicts of interest to report regarding the present study.

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