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Integrated GC-MS and LC-MS-Based Untargeted Metabolomics Reveals Diverse Metabolites in Fermented Pine Needles

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ABSTRACT

The diversity of metabolites produced in fermented pine needles at different stages of fermentation has rarely been investigated. In the present study, untargeted metabolomic analysis using GC-MS and LC-MS was performed to detect metabolites in fermented pine needles at different fermentation stages. A total of 30 samples of pine needles fermented first with yeast (*Saccharomyces cerevisiae*) and then with a mixed bacterial culture of *Lactobacillus fermentum* CECT5716 and *Bifidobacterium Breve* M16V, were used to detect differential metabolites at different stages of fermentation. Pearson's correlation analysis was used to determine correlations between metabolites and key microbial communities. A total of 708 differential metabolites (430 from LC-MS and 278 from GC-MS analysis) were identified. The PCA and OPLS-DA revealed distinct differences between metabolites at different fermentation stages. Key differential metabolites identified through GC-MS analysis included; Phosphoric acid, D-Fructose, 2-O-alpha-mannosyl-D-glycerate, 1,3-dihydroxyacetone dimer, Galactosylglycerol, 2-Isopropylmalic acid, alpha-D-Galactose, Citrate, 4-Hydroxycinnamic acid, and Shikimate. Similarly, key differential metabolites identified through LC-MS included; 2-Phenylethanol, Dimethylglycine, 2-Hydroxybenzaldehyde, 3-Aminoisobutanoic acid, p-Cresol, Triethylamine, 2-Ketobutyric acid, Cytosine, Benzaldehyde, and Creatinine. Annotation of differential metabolites to KEGG pathway enrichment analysis revealed the association of these metabolites with phenylpropanoid, flavonoid, and secondary metabolite biosynthesis. Furthermore, the results showed that three bacterial (*Firmicutes*, *Actinobacteria*, and *Lactobacillus*) and three fungal genera (*Penicillium*, *Candida*, and *Basidiomycota*) significantly correlated with differential metabolites showing synergistic effects. Our study reveals a comprehensive comparison of metabolites at different fermentation stages and provides practical insights into the mechanism of metabolite enrichment in fermented pine needles.

KEYWORDS

Fermented pine needle; *Pinus massoniana*; GC-MS; LC-MS; metabolomics

1 Introduction

Pinus massoniana is a species of coniferous evergreen needle-leaved trees (family Pinaceae), widely distributed in East Asia including Japan, Korea, North-Eastern China, and South-East Russia [1]. Various parts of the pine tree, such as pine needles, cones, cortex, and pollen, have been used as folk medicine or food [2,3]. Pine needles contain diverse phytochemical compounds including phenolics, cinnamic acids,



terpenoids, carboxylic acids, fatty acids, and flavonoids [4]. Owing to these bioactive compounds, pine needles possess a wide array of biological activities such as antimicrobial, antimutagenic, antioxidant, anti-neurodegenerative, and cardioprotective properties [5–7]. In addition to secondary metabolites, pine needles also contain some significant minerals (calcium 28 mg/100 g), amino acids (glutamic acid, lysine, phenylalanine, and leucine), and vitamins (niacin, beta carotene, riboflavin, and thiamine) [8]. Despite of rich nutrient profile and bioactive compounds, substantially high concentration of condensed tannins and antinutritional factors in pine needles may interfere with nutrient absorption and protein digestibility in the gut [9]. In this regard, fermentation with probiotics is one of the best ways to remove antinutritional factors (e.g., tannins) and improve the flavor, palatability, digestibility, and nutrient availability of pine needles [10,11]. Fermentation can also enhance the shelf life of food while improving biochemical activities like antioxidant and antibacterial properties [11].

Fermented pine needle drink has been utilized as a functional beverage in the countries of Asia-Pacific region particularly Taiwan, Korea, and China owing to its various biological properties. Studies have reported that fermented pine needle extract is rich in phenolics, essential oils, flavonoids, and terpenoids, with potential health-promoting potential [1,10]. Self-fermented pine needle extract has been shown to decrease blood lipid levels and regulate gastrointestinal motility [1]. Additionally, pine needles are used as a health-improving agent in teas, extracts, and some alcoholic beverages [12]. Ethanolic extracts of pine needles have also shown antidepressant activities and hence can be used to amend unidentified clinical symptoms, such as fatigue, anxiety, depression, and sleep disorders [13–15]. Moreover, the potent antioxidant activities of pine needles have been widely reported which are responsible for their excellent free radical scavenging potential and cytotoxicity against cancer cells [8,16]. These antioxidant properties of pine needles are mainly attributed to their rich content of proanthocyanidins, catechins, and essential oils, which are considered to be potent antioxidants and thus may play an important role as anti-aging compounds and in the prohibiting of chronic diseases including cardiovascular disorders, cancer, diabetes, and hypertension [11,17,18]. Therefore, fermented pine needles have been traditionally processed and used to treat a variety of diseases.

It is worth mentioning that pine needles have gained much attention in recent years for their use as an ingredient in herbal teas and various food recipes, owing to their potential health benefits [18]. To better exploit the potential of pine needles as a functional food and its nutraceutical value, it is imperative to comprehensively analyze the diversity of metabolites of fermented pine needles. Various fermentation conditions including probiotic strains, temperature, and duration of incubation substantially affect the quantity and quality of metabolites in pine needles. Hence metabolomic study of fermented pine needles is essentially required to comprehensively analyze metabolites at a given time or under given conditions of fermentation [19]. Spectral analysis of metabolites can be performed by gas chromatography or liquid chromatography combined with mass spectrometry (GC-MS and LC-MS, respectively) [20]. In this regard, nontargeted metabolomic analysis is a desirable approach to get as much information as possible from biological samples like fermented pine needles [21]. Recently, a study has determined the total phenols and flavonoid content of pine needles fermented with *Lactobacillus plantarum* SK4315, *Saccharomyces cerevisiae*, and their co-culture through ELISA quantification [19]. But still, no study has performed metabolomic characterization of fermented pine needles to identify individual compounds which is essentially required to understand their potential health benefits and potential applications.

Therefore, in the present study, we evaluated the variation in metabolites at different stages of the fermentation process of pine needles through untargeted GC-MS and LC-MS analysis. Additionally, the correlation of metabolites with the microbiome was also investigated. The present study envisioned to provide novel insights into the mechanism of improvement in the quality of fermented pine needles and their potential utility for developing traditional fermented foods.

2 Materials and Methods

2.1 Preparation of Pine Needle Fermentation Extract

Pine needle fermentation extract was obtained as described in our previous study [22]. Briefly, the fresh pine needles were collected from pine trees (*Pinus massoniana*) at Cha'en Temple, Shaoshan, China. The *Pinus massoniana* trees in Shaoshan's Cha'en Temple were harvested to collect fresh pine needles. The needles were ground to a fineness of 0.5 to 1.2 mm and air-dried for three days in dry shade. This was followed by a 1:10 ratio of the needles to water. The mixture was added and evenly dissolved, consisting of 22% brown sugar and 33% rock sugar. Finally, 0.8% active dry yeast was added, and the mixture was left to incubate for 18 days at 22°C. After that, a mixed bacterial solution containing 6% *Lactobacillus plantarum* and 1.0:1.5 Bifidobacterium Breve M16V was added to inoculate it. Following 7 days at 37°C, the fermentation process was shifted to room temperature (about 25°C) and kept for a total of 180 days. The chemicals used in the creation of the extract were purchased by Tianjin Jinteng Experimental Equipment Co., Ltd. (Tianjin) in China. The chemicals included glucose and bovine serum albumin, both of which had purities of more than 98%. Shanghai Aladdin Biochemical Technology Co., Ltd., Shanghai, China provided the 2-Chloro-L-phenylalanine (internal standard material), d₃H₂O (Millipore, St Louis, MO, USA), and methanol (purity > 98%, Thermo Fisher, Waltham, MA, USA).

2.2 Sampling of Fermented Pine Needle Extract

Starting from day 0 of fermentation with unfermented pine needles, samples of pine needle fermentation juice were collected regularly on day 1 (GD1), 3 (GD3), 9 (GD9), and 18 (GD18) of yeast fermentation and after mixed bacterial fermentation at day 2nd (HD2) and 7th (HD7) for metabolite detection through GC-MS analysis. The same samples were also used for the detection of metabolites through LC-MS analysis (coded as LD1, LD3, LD9, and LD18 for yeast fermentation and LHD2 and LHD7 for mixed bacterial fermentation). These samples were stored at -80°C until used for further processing. To ensure the reproducibility of the experiment, approximately 5 kg of pine needles were loaded into each fermentation tank. During the fermentation process, a total of 5 parallel groups were set up, and 3 samples were randomly selected for analysis. A total of 5 samples were analyzed for each fermentation stage.

2.3 LC-MS Analysis

The pine needle fermentation extract was thawed, and then well mixed for one minute before being transferred to a two-milliliter centrifuge tube and concentrated until it was dry. Subsequently, 500 µL of methanol (stored at -20°C) was added to the sample tube, vortexed for 1 min, and centrifuged at 12,000 rpm at 4°C for 10 min. The sample was then concentrated until it was dry after the whole supernatant was transferred to a fresh 2 mL centrifuge tube. The dried material was then dissolved in 150 µL of a 4-ppm solution of 2-chloro-L-phenylalanine made with 80% methanol, and the resulting supernatant was then passed through a 0.22 µm membrane filter. For LC-MS detection, the filtrate was subsequently introduced to the detection bottle. A Vanquish UHPLC System (Thermo Fisher Scientific, Waltham, MA, USA) and an ACQUITY UPLC® HSS T3 (150 mm × 2.1 mm, 1.8 µm) column (Waters, Milford, MA, USA) were used for the LC-MS analysis. At a temperature of 40°C, a 2 µL sample was injected into the column at a flow rate of 0.25 mL/min. The mobile phase in the positive ion mode was made up of 0.1% formic acid in acetonitrile (B2) and 0.1% formic acid in water (A2). The gradient elution procedure was used as reported previously [23]. Using an electrospray ion source (ESI)-equipped Q Exactive Focus equipment (Thermo Fisher Scientific), mass spectrometry was used to identify metabolites. A capillary temperature of 325°C, a sheath gas flow rate of 30 arb, and an auxiliary gas flow rate of 10 arb were the additional parameters. Full-scan MS analysis (m/z 100–1000) yielded a resolution of 70,000 FWHM for all metabolites. The resolving power for dd-MS2 was configured at

17,500 FWHM. A collision energy of 30 eV was used for collision-induced dissociation (CID), and superfluous MS/MS information was eliminated via dynamic exclusion [24].

2.4 GC-MS Analysis

About 100 μL of the sample and 1 mL of solution (containing acetonitrile, isopropanol, and water at 3:3:3:2 v/v/v) were combined in a 2 mL EP tube (-20°C) to analyze the metabolites in the pine needle extract. After 30 s of vortex, the mixture was exposed to ultrasonication for 5 min. After centrifuging the mixture for two minutes at 12,000 rpm, 500 μL of the supernatant was taken in a 2 mL EP tube. After that, the sample was concentrated under vacuum until it was totally dry, which usually took eight to ten hours. Subsequently, a 20 mg/mL methoxy pyridine solution was added to 80 μL of the dried and concentrated sample, stirred for 30 s, and incubated for 60 min at 60°C . After 30 s of mixing, 100 μL of BSTFA-TMCS (99:1) derivatization reagent was incubated at 70°C for 90 min. After that, the mixture was centrifuged for three min at 14,000 rpm. Approximately 90 to 100 μL of the supernatant was transferred to the detection vial following centrifugation. For testing, the material was kept in a sealed cup, and the Gas Chromatography time-of-flight detection was finished in a day.

To separate the derivatives, gas chromatography was performed on DB-5MS capillary column (Agilent J&W Scientific, Folsom, CA, USA, 30 m \times 250 μm i.d., 0.25 μm film thickness), with a constant-flow (1 mL/min) of helium. Using an automated injector with a split ratio of 1:10, 1 μL of sample was injected. Using an automated injector with a split ratio of 1:10, 1 μL of sample was injected. The transmission line and ion source temperatures were used as reported previously [25].

2.5 Data Processing and Multivariate Analysis

The original raw data was converted to mzXML format using MSConvert in the ProteoWizard program (v3.0.8789) [26]. After that, it was processed using XCMS [27] for features, alignment, and retention time correction. Metabolites were identified based on precise molecular mass and MS/MS data by comparing with HMDB [28], massbank [29], LipidMaps [30], mzcloud [31], and KEGG [32] databases. Robust LOESS signal correction (QC-RLSC) [24] was employed for data normalization to address any systematic bias. To ensure precise metabolite identification, only ion peaks with relative standard deviations in quality control less than 30% were retained after normalization. Metabolome data were analyzed by the ropis R package as reported previously [33]. MetaboAnalyst was used to identify differential metabolites [34]. To improve the overall understanding of the detected metabolites, pathway enrichment, and topological studies were integrated. The KEGG pathway was then mapped to the detected metabolites, which made it easier to comprehend higher-level systemic biological activities. Using the KEGG Mapper program, the metabolites and associated pathways were visualized.

Differential metabolites were defined as those that matched the following criteria: $\text{VIP} > 1$, $p\text{-value} < 0.05$, fold change ≥ 2 , or $\text{FC} \leq 0.5$. When the ratio criteria of $x/n > y/N$ is met, it indicates enrichment in the metabolic pathway enrichment analysis. Additionally, if the $p\text{-value}$ was < 0.05 of a metabolic pathway, it was considered enriched statistically.

Pearson correlation was used to elucidate the relationship between metabolites and microbiome as described in our previous study [22]. The heatmap was employed to show significant positive or negative correlations between metabolites and microbiome. Data of major bacterial and fungal genera were employed from our companion study to perform correlation [22]. The correlation between differential metabolites and microbial communities was analyzed by employing the cor function in R version 4.0.3 (R Project for Statistical Computing, Vienna, Austria). The significance of the correlation between differential metabolites was determined by cor. Mtest() function and plotted by corrplot package of R. The $p\text{-value}$ of less than 0.05 was considered as the statistical significance.

3 Results

3.1 Untargeted Metabolomic Analysis

The base peak chromatogram of negative ion mode (ESI⁻), and positive ion mode (ESI⁺), in LC/MS metabolite profiles are shown in Fig. S1. The unsupervised principal component analysis (PCA) was employed to evaluate the quality of the GC/MS and LC/MS data. The distribution of metabolic profiles for the test and QC samples in PCA is shown in Fig. 1. Results of PCA indicated overall good reproducibility of results. A clear group separation between GD1, GD3, and other fermentation stages was observed in the PCA score plots (Fig. 1). The PC1 and PC2 explained 46.4% and 11.6% of the total variance in ESI⁻ mode, while 45.2% and 13.3% of the total variance of all samples in ESI⁺ mode, respectively.

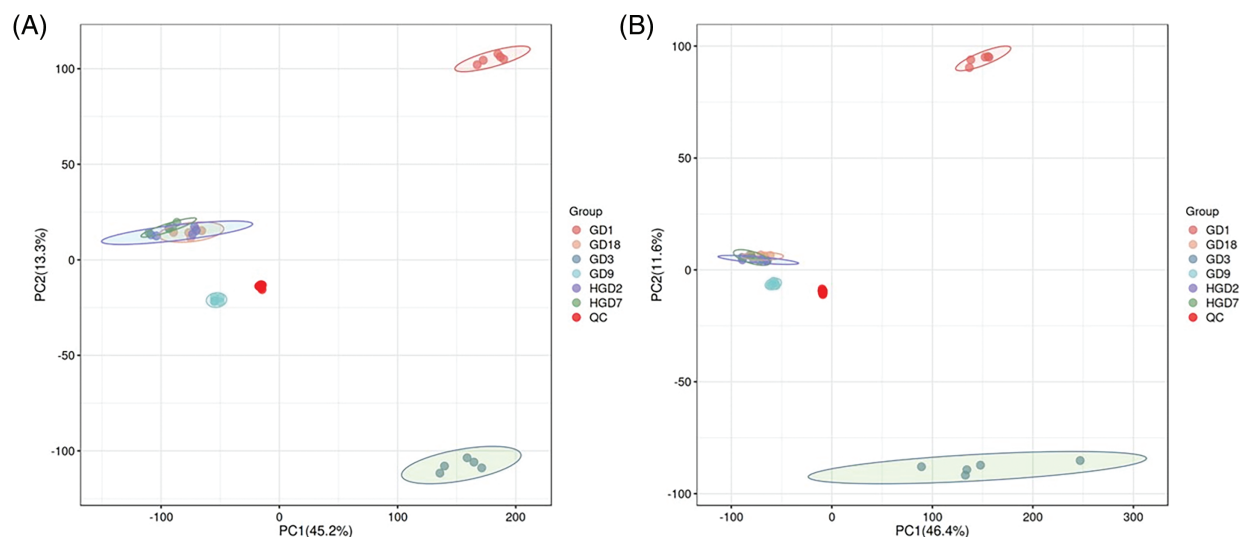


Figure 1: The PCA score scatter plot under positive ion mode (A) and negative ion mode (B)

3.2 Identification of Differential Metabolites

A supervised OPLS-DA was conducted to identify group variables. The OPLS-DA clearly separated GD1, GD3, and other fermentation stages (Fig. 2). We used $VIP > 1$ and $p\text{-value} < 0.05$ to identify differential metabolites between different fermentation stages. The results revealed a total of 708 differential metabolites identified by LC-MS (430) and GC-MS (278) analyses (Tables 1 and 2). In addition, differential metabolites identified through LC-MS and GC-MS at different stages of fermentation of pine needles are shown in Fig. S2A,B, respectively.

3.3 Metabolic Pathway of Differentially Abundant Metabolites

The MetaboAnalyst 4.0 was used to identify the pathways for differential metabolites among different fermentation stages through KEGG pathway analysis. All the different metabolites were annotated to the biological pathways listed in the KEGG database (Fig. 3). These differentially enriched pathways are grouped into phenylpropanoid biosynthesis, flavonoid biosynthesis, and secondary metabolite biosynthesis. Results revealed seven pathways with impact values greater than 0.1 (threshold value for enrichment and topology analysis) among these differentially enriched pathways.

3.4 Association of Metabolites with Microbial Communities in Fermented Pine Needles

Major differential metabolites with high VIP values (top 20) and key abundant bacteria and fungi genera (top 10) observed in fermented pine needles were selected to determine the correlation between microbial

communities and metabolites (Fig. 4). Results revealed that Firmicutes, Lactobacillus, and Actinobacteria positively correlated ($p < 0.05$) with dimethylglycine, 2-ketobutyric acid, cyto-sine, 2-O-alpha-mannosyl-D-glycerate and 1,3-dihydroxyacetone dimer but negatively correlated ($p < 0.05$) with phosphoric acid. However, Proteobacteria negatively correlated with galactosylglycerol, 1,3-dihydroxyacetone dimer, 2-O-alpha-mannosyl-D-glycerate, dimethylglycine, 2-phenylethanol, creatinine, triethylamine, cytosine, and 2-ketobutyric acid, while positively correlated with phosphoric acid. Lactobacillus was positively associated with dimethylglycine, 2-ketobutyric acid, cytosine, 1,3-dihydroxyacetone dimer, and 2-O-alpha-mannosyl-D-glycerate but negatively correlated with phosphoric acid. Regarding fungal taxa, three fungal genera including Penicillium, Candida and Basidiomycota exhibited a negative correlation ($p < 0.05$) with the top ten metabolites identified with LC-MS while Schizoasaccharomyces and Ascomycota showed a positive correlation with these metabolites (Fig. 4D). However, Penicillium, Candida, and Basidiomycota exhibited a negative correlation with galactosylglycerol, 1,3-dihydroxyacetone dimer, 2-O-alpha-mannosyl-D-glycerate, dimethylglycine, 2-phenylethanol, creatinine, triethylamine, cytosine, and 2-ketobutyric acid, while positively correlated with phosphoric acid. Both Schizoasaccharomyces and Ascomycota positively correlated with almost all top 10 metabolites (identified through GC-MS) but negatively correlated with D-Fructose and Phosphoric acid.

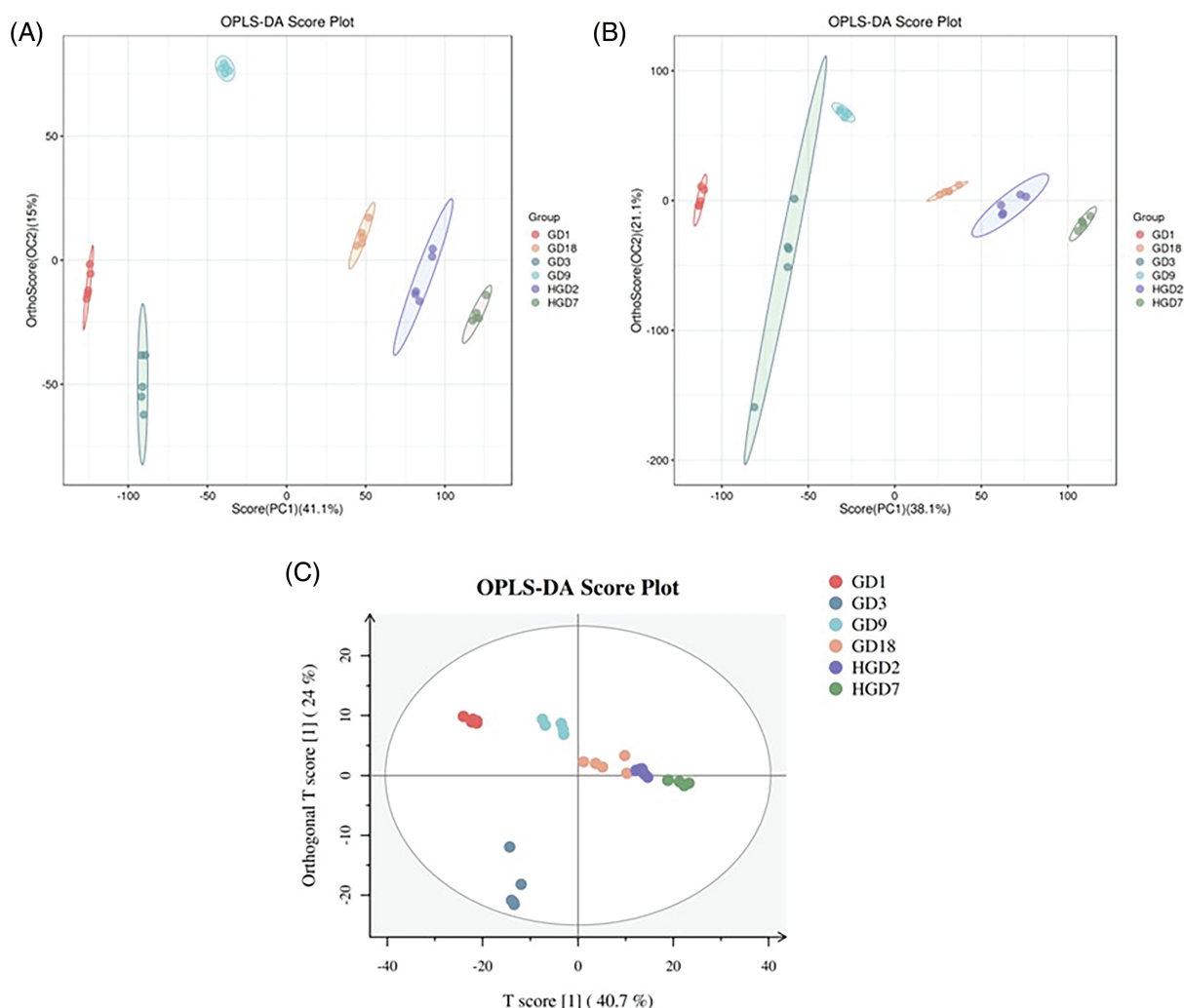


Figure 2: The OPLS-DA score scatter plot under positive ion mode (A) and negative ion mode (B) and OPLS-DA score plot (C)

Table 1: Differentially abundant metabolites identified by LC-MS in different pine needle groups

Group	Name	Formula	VIP	<i>p</i> -value	Regulation
LD1 vs. LD3	2-Ketobutyric acid	C ₄ H ₆ O ₃	1.337789	0.007937	Down
	2-Phenylethanol	C ₈ H ₁₀ O	1.217314	0.007937	Up
	o-Xylene	C ₈ H ₁₀	1.202172	0.007937	Up
	Benzaldehyde	C ₇ H ₆ O	1.243078	0.007937	Up
	(S)-1-Pyrroline-5-carboxylate	C ₅ H ₇ NO ₂	1.228277	0.007937	Up
	2-Heptanone	C ₇ H ₁₄ O	1.059129	0.007937	Down
	Dihydrouacil	C ₄ H ₆ N ₂ O ₂	1.304614	0.007937	Down
	L-2,4-diaminobutyric acid	C ₄ H ₁₀ N ₂ O ₂	1.263681	0.007937	Up
	Betaine	C ₅ H ₁₁ NO ₂	1.131496	0.007937	Down
	Benzeneacetonitrile	C ₈ H ₇ N	1.194636	0.007937	Down
	Triethylamine	C ₆ H ₁₅ N	1.201654	0.007937	Down
	2-Ketobutyric acid	C ₄ H ₆ O ₃	1.029964	0.007937	Down
LD1 vs. LD18	Benzonitrile	C ₇ H ₅ N	1.011184	0.007937	Down
	3-Aminoisobutanoic acid	C ₄ H ₉ NO ₂	1.181723	0.007937	Down
	Dimethylglycine	C ₄ H ₉ NO ₂	1.201479	0.007937	Down
	2-Phenylethanol	C ₈ H ₁₀ O	1.191934	0.007937	Down
	Benzaldehyde	C ₇ H ₆ O	1.19113	0.007937	Down
	p-Cresol	C ₇ H ₈ O	1.108833	0.007937	Down
	Cytosine	C ₄ H ₅ N ₃ O	1.039186	0.007937	Down
	Creatinine	C ₄ H ₇ N ₃ O	1.016412	0.007937	Down
	Triethylamine	C ₆ H ₁₅ N	1.184234	0.007937	Down
	2-Ketobutyric acid	C ₄ H ₆ O ₃	1.159711	0.007937	Down
	Benzonitrile	C ₇ H ₅ N	1.07846	0.007937	Down
	3-Aminoisobutanoic acid	C ₄ H ₉ NO ₂	1.182985	0.007937	Down
LD1 vs. HLD7	Dimethylglycine	C ₄ H ₉ NO ₂	1.185183	0.007937	Down
	2-Phenylethanol	C ₈ H ₁₀ O	1.176857	0.007937	Down
	Benzaldehyde	C ₇ H ₆ O	1.037843	0.007937	Down
	p-Cresol	C ₇ H ₈ O	1.110365	0.007937	Down
	Cytosine	C ₄ H ₅ N ₃ O	1.035434	0.007937	Down
	Creatinine	C ₄ H ₇ N ₃ O	1.050803	0.007937	Down
	Triethylamine	C ₆ H ₁₅ N	1.166145	0.007937	Down
	2-Ketobutyric acid	C ₄ H ₆ O ₃	1.233262	0.007937	Down
	3-Aminoisobutanoic acid	C ₄ H ₉ NO ₂	1.234114	0.007937	Down
	Dimethylglycine	C ₄ H ₉ NO ₂	1.276917	0.007937	Down
	2-Phenylethanol	C ₈ H ₁₀ O	1.276342	0.007937	Down
	o-Xylene	C ₈ H ₁₀	1.074885	0.007937	Down

(Continued)

Table 1 (continued)						
Group	Name	Formula	VIP	<i>p</i> -value	Regulation	
LD3 vs. LD9	Benzaldehyde	C ₇ H ₆ O	1.20529	0.007937	Down	
	p-Cresol	C ₇ H ₈ O	1.220418	0.007937	Down	
	Cytosine	C ₄ H ₅ N ₃ O	1.044532	0.007937	Down	
	Creatinine	C ₄ H ₇ N ₃ O	1.248099	0.007937	Down	
	Triethylamine	C ₆ H ₁₅ N	1.608799	0.007937	Up	
	p-Cresol	C ₇ H ₈ O	1.572139	0.007937	Down	
	Phenylacetaldehyde	C ₈ H ₈ O	1.090374	0.007937	Down	
	2-Hydroxybenzaldehyde	C ₇ H ₆ O ₂	1.615554	0.007937	Down	
LD9 vs. LD18	4-Methylbenzyl alcohol	C ₈ H ₁₀ O	1.394152	0.007937	Up	
	Furan-2,5-dicarbaldehyde	C ₆ H ₄ O ₃	1.495748	0.007937	Up	
	1,2,3-Trihydroxybenzene	C ₆ H ₆ O ₃	1.547897	0.007937	Up	
	Adipate semialdehyde	C ₆ H ₁₀ O ₃	1.526985	0.007937	Up	
	5,6-Dihydro-5-fluorouracil	C ₄ H ₅ FN ₂ O ₂	1.620015	0.007937	Down	
	(2R,4S)-2,4-Diaminopentanoate	C ₅ H ₁₂ N ₂ O ₂	1.514634	0.007937	Down	
	2-Hydroxybenzaldehyde	C ₇ H ₆ O ₂	2.40485	0.007937	Down	
	para-Cresidine	C ₈ H ₁₁ NO	2.567277	0.007937	Down	
	Hordenine	C ₁₀ H ₁₅ NO	2.392332	0.007937	Down	
	Diphenylamine	C ₁₂ H ₁₁ N	2.071168	0.007937	Down	
LD18 vs. HLD2	Glucosamine	C ₆ H ₁₃ NO ₅	2.373594	0.007937	Down	
	Ecgonine methyl ester	C ₁₀ H ₁₇ NO ₃	2.127741	0.007937	Down	
	Antiarol	C ₉ H ₁₂ O ₄	2.271842	0.007937	Down	
	N-Benzyloxycarbonylglycine	C ₁₀ H ₁₁ NO ₄	2.280004	0.007937	Down	
	Zerumbone	C ₁₅ H ₂₂ O	2.324038	0.007937	Up	
	Carnosine	C ₉ H ₁₄ N ₄ O ₃	2.217233	0.007937	Down	
	Niacinamide	C ₆ H ₆ N ₂ O	1.591885	0.007937	Down	
	2-Hydroxybenzaldehyde	C ₇ H ₆ O ₂	2.289874	0.007937	Down	
	4-Hydroxybenzyl alcohol	C ₇ H ₈ O ₂	2.109329	0.007937	Down	
	5,6-Dihydro-5-fluorouracil	C ₄ H ₅ FN ₂ O ₂	2.27334	0.007937	Down	
	Anabasine	C ₁₀ H ₁₄ N ₂	2.549552	0.007937	Down	
	HLD2 vs. HLD7	Hordenine	C ₁₀ H ₁₅ NO	2.229305	0.007937	Down
		Phosphohydroxypyruvic acid	C ₃ H ₅ O ₇ P	2.237604	0.007937	Down
Epicatechin		C ₁₅ H ₁₄ O ₆	1.827618	0.007937	Down	
12(S)-HpETE		C ₂₀ H ₃₂ O ₄	2.445905	0.007937	Down	
Prostaglandin F _{2b}		C ₂₀ H ₃₄ O ₅	2.640548	0.007937	Up	

Note: LD1, LD3, and LD9 = pine needles fermented with yeast for 1, 3, and 9 days, respectively; LD18 = pine needles fermented with mixed bacteria culture; HLD2 and HLD7 = post-fermentation after 2 days and 7 days, respectively.

Table 2: Identification of differentially abundant metabolites through GC-MS between the different groups

Treatment group	Metabolite	VIP	<i>p</i> -value	Regulation
GD1 vs. GD3	Phosphate	1.127243	0.012186	Up
	Urea	1.201585	0.012186	Up
	Quinic acid	2.52315	0.012186	Up
	trans-4-Hydroxycyclohexanecarboxylic acid	1.361799	0.012186	Up
	Shikimic acid	1.381419	0.012186	Up
	Citrate	3.045124	0.012186	Up
	Diisopropanolamine	1.235646	0.012186	Up
	D-Mannitol	1.287403	0.012186	Down
	L-Iditol	1.767427	0.012186	Down
	D-Mannonate	1.844432	0.012186	Down
	Phosphoric acid	1.405633	0.012186	Up
GD3 vs. GD9	Sulphuric acid	1.513336	0.012186	Up
	Urea	1.374769	0.012186	Down
	Galactosylglycerol	1.492195	0.012186	Down
	2-O-alpha-Mannosyl-D-glycerate	1.4862	0.012186	Down
	4-Hydroxyphenyllactic acid	1.770848	0.012186	Down
	d-Glycero-d-galacto-heptose	1.857868	0.012186	Down
	D-Pinitol	1.139059	0.012186	Up
	2-Isopropylmalic acid	1.267486	0.012186	Down
	Citrate	1.49763	0.012186	Down
	2-O-alpha-Mannosyl-D-glycerate	1.673238	0.012186	Up
	Uracil, 1-beta-D-ribofuranosyl-	1.060592	0.012186	Down
GD9 vs. GD18	4-Hydroxyphenyllactic acid	2.229733	0.012186	Down
	Phloretate	2.36933	0.012186	Down
	Vanillic acid	1.056876	0.012186	Down
	Phenylethyl alcohol	1.994029	0.012186	Down
	d-Glycero-d-galacto-heptose	3.047591	0.012186	Up
	Quinic acid	2.197517	0.012186	Up
	D-Glucaro-1,4-lactone	1.584852	0.012186	Up
	cis-Aconitic acid	1.45302	0.012186	Down
	Galactosylglycerol	1.670699	0.012186	Up
	4-Hydroxyphenyllactic acid	2.098989	0.012186	Up
	Citrate	1.749682	0.012186	Up
D-Fructose	4.230729	0.012186	Down	

(Continued)

Table 2 (continued)				
Treatment group	Metabolite	VIP	<i>p</i> -value	Regulation
GD18 vs. HGD2	D-(-)-Fructose	3.573803	0.012186	Down
	beta-D-Fructose	2.059645	0.012186	Up
	alpha-D-Galactose	2.696735	0.012186	Down
	Allose	1.536538	0.012186	Up
	D-Glucuronic acid	1.444575	0.012186	Down
	Galacturonic acid	1.422559	0.012186	Down
	4-Hydroxyphenyllactic acid	2.127418	0.012186	Down
	1,3-Dihydroxyacetone dimer	3.512746	0.012186	Down
	D-Fructose	4.7469	0.012186	Up
	D-(-)-Fructose	3.884975	0.012186	Up
HGD2 vs. HGD7	beta-D-Fructose	2.227115	0.012186	Down
	alpha-D-Galactose	2.383098	0.012186	Up
	Allose	1.321844	0.012186	Down
	L-Pipecolate	1.042954	0.012186	Down
	L-Gulonolactone	1.09369	0.012186	Up
	DL-Alanine	3.248182	0.012186	Down
	Phosphoric acid	1.642315	0.012186	Up
	Galactosylglycerol	1.320631	0.012186	Down
	2-O-alpha-Mannosyl-D-glycerate	1.063338	0.012186	Down
	4-Hydroxyphenyllactic acid	2.213768	0.012186	Down
GD1 vs. GD18	Phloretate	1.470491	0.012186	Down
	Phenylethyl alcohol	1.199215	0.012186	Down
	D-Pinitol	1.35218	0.012186	Up
	Quinic acid	2.356941	0.012186	Up
	2-Isopropylmalic acid	1.258468	0.012186	Down
	Shikimate	1.907375	0.012186	Up
	Phosphoric acid	1.623491	0.012186	Up
	Urea	1.404209	0.012186	Down
	Galactosylglycerol	1.065492	0.012186	Down
	4-Hydroxyphenyllactic acid	2.179253	0.012186	Down
GD1 vs. HGD7	Phloretate	1.469468	0.012186	Down
	D-Pinitol	1.337966	0.012186	Up
	Quinic acid	2.779405	0.012186	Up
	2-Isopropylmalic acid	1.198079	0.012186	Down
	Shikimate	1.791044	0.012186	Up
	Shikimic acid	1.121891	0.012186	Up

Note: GD1, GD3, and GD9 = pine needles fermented with yeast for 1, 3, and 9 days, respectively; GD18 = pine needles fermented with mixed bacteria culture; HD2 and HD7 = post-fermentation after 2 days and 7 days, respectively.

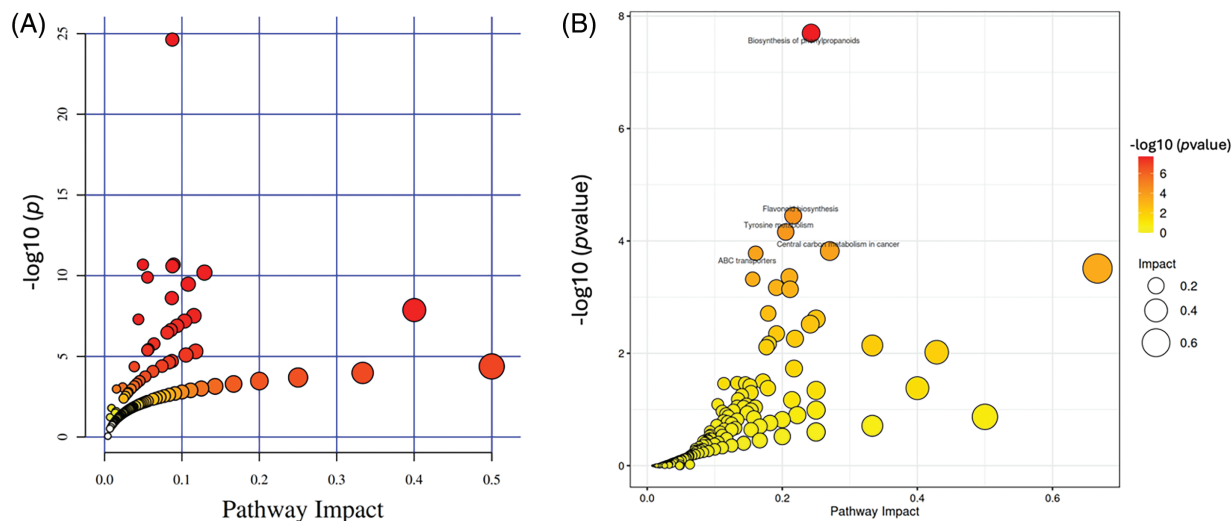


Figure 3: Summary of pathway analysis with MetaboAnalyst. (A) LC-MS; (B) GC-MS

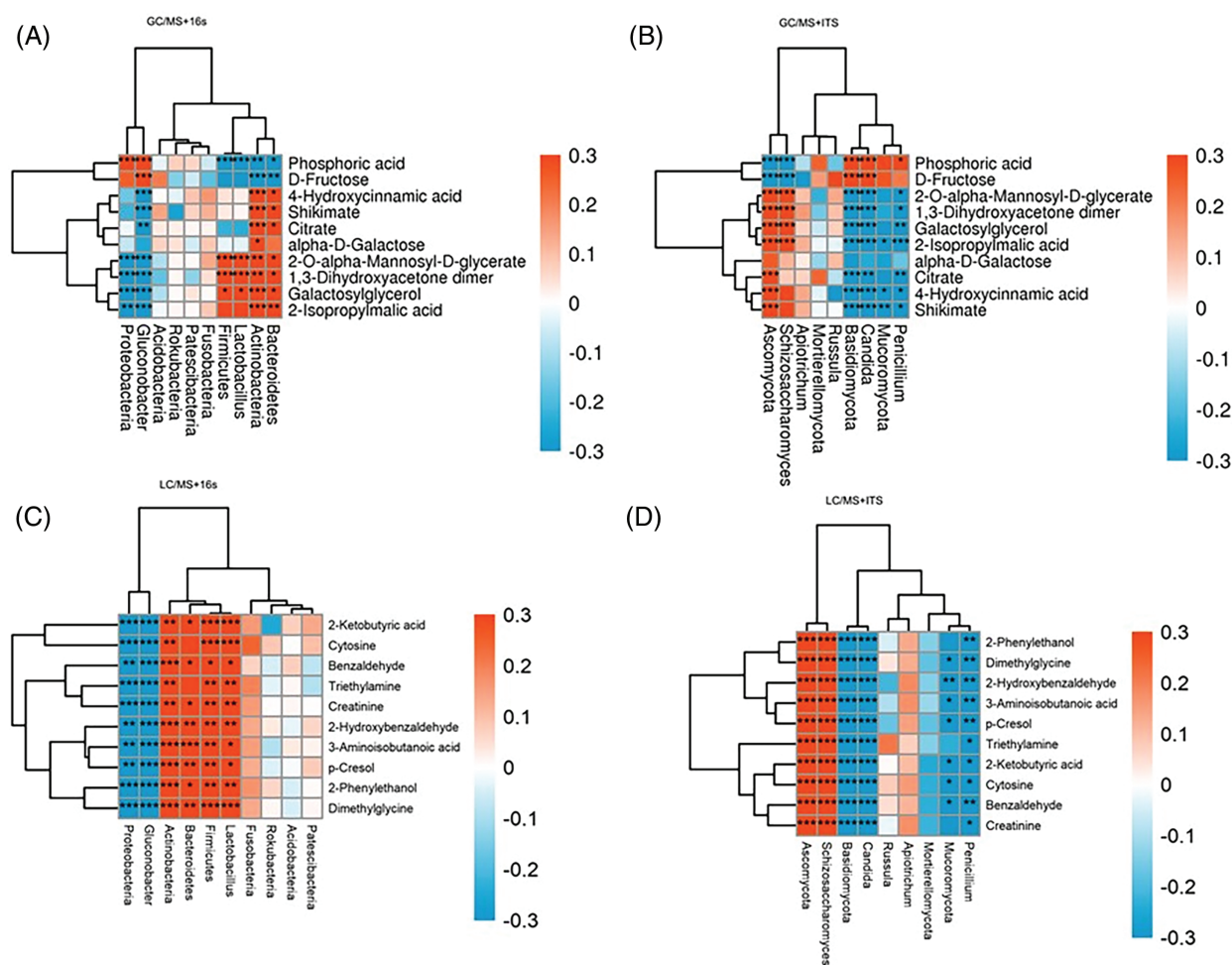


Figure 4: Correlation of bacterial (A and C) and fungi (B and D) with key metabolites of fermented pine needles identified through GC-MS and LC-MS analysis, respectively

4 Discussion

The untargeted metabolomic analysis (through LC-MS and GC-MS) of fermented pine needles was conducted to detect a wider range of metabolites because each analytical platform can analyze different types of metabolites. Wider use of GC-MS for metabolomic analysis compared to LC-MS is attributed to available metabolite databases [31]. However, both these analyses are preferred for metabolomic analysis. In the present study, untargeted metabolomic analysis of fermented pine needles successfully detected 207 metabolites by GC-MS, and 136 metabolites by LC-MS analysis.

Differential metabolites observed in the present study are attributed to the fact that fermentation time and inoculant can greatly affect the metabolite contents as more polar phytochemicals are released during the fermentation process [19]. Moreover, different microbial strains produce diverse metabolites during fermentation depending upon the availability of respective substrates. Fermentation of pine needles with *Bacillus subtilis* exhibited metabolite contents rich in phenolics and flavonoids produced through various enzymatic reactions that triggered the synthesis of new phenolic compounds [10,35]. Similarly, stronger antioxidant activity for *P. morrisonicola* and its fermented products has been reported compared to its non-fermented counterparts owing to its rich metabolites including phenolic compounds, flavonoids, essential oils, and terpenoids. Even self-fermented pine needles have shown potential bioactivities to address hypercholesterolemia and its associated conditions owing to their rich metabolites [1,36]. A diverse range of metabolites was identified at different fermentation stages of fermented pine needles, with some metabolites having potential health-promoting effects and economic significance. These findings are in agreement with a recent study reporting a substantial increase in total polyphenolic and flavonoid contents after the fermentation of pine needles with *Lactobacillus plantarum* SK4315 fermentation and *Saccharomyces cerevisiae* SK3587 leading to higher antioxidant, antimicrobial, and enzymatic activities compared to natural fermented group [19]. It is evident from various studies that metabolite contents and biological activities largely depend on the type of fermentation, fermentation time, and inoculant owing to the enrichment of metabolite contents [11,35].

The present study revealed that differentially enriched metabolites were observed in different fermentation groups. For instance, an aromatic alcohol 2-phenylethanol (2-PE) was found upregulated in yeast fermented pine needles but it decreased with the passage of fermentation time. This compound has a characteristic rose-like odor and is an important raw material in the cosmetics, perfumery, and food industries. Smaller amounts of 2-PE are used in the food industry to enhance the taste of soft drinks, candy, cookies, and other foods [37]. Nevertheless, 2-PE is an important aroma compound found in low concentrations in fermented foods and beverages, such as cheese and wine. It is also reported that 2-PE produced by *Pichia anomala* can inhibit the growth of *Aspergillus flavus* [38,39]. In our study, 2-PE contents significantly differed among the different fermentation groups (LD1 vs. LD3, LD1 vs. LD18, LD1 vs. HLD7, LD3 vs. LD9) as its synthesis was downregulated with the passage of fermentation. Proteobacteria, Candida, and Basidiomycota negatively correlated with 2-PE contents, which may be the main reason that influences the fragrance in the different fermentation stages of fermented pine needles. The present study revealed that synthesis of Shikimic acid and Shikimate was upregulated during the fermentation as it positively correlated with Bacteroides and Actinobacteria while negatively correlated with Gluconobacter (Proteobacteria). Shikimic acid is essential to the health of gut bacteria and has been shown to modulate microbiota and NF- κ B/MAPK signaling pathway to impart beneficial effects on intestinal integrity and inflammatory conditions [40,41]. Hence, the Shikimic contents of fermented pine needles are crucial for their positive health benefits.

The findings of the present study demonstrated that Lactobacillus, Firmicutes, Bacteroidetes and Actinobacteria positively correlated with key metabolites including 2-Phenylethanol, Dimethylglycine,

2-Hydroxybenzaldehyde, 3-Aminoisobutanoic acid, p-Cresol, Triethylamine, 2-Ketobutyric acid, Cytosine, Benzaldehyde, and Creatinine. It means that fermentation with *Lactobacillus* along with other bacteria can increase the contents of major metabolites leading to major bioactivities as reported earlier [1,10]. Moreover, these bacteria revealed a negative association with D-Fructose and Phosphoric acid. *Lactobacillus* degrades carbohydrates to produce lactic acid which is a precursor for 3-hydroxy-2-butanone and other acids. Additionally, owing to their strong fermentation ability of *Lactobacillus*, it derives volatile and non-volatile flavor contents with the release and degradation of free amino acids in fermented products. The amino acid transferases produced by *Lactobacillus*, can convert phenylalanine to phenylacetaldehyde and benzaldehyde; and convert leucine to generate 3-methyl-1-butanol, etc. These metabolites are major flavor compounds produced during the fermentation of pine needles. Moreover, controlled fermentation with *Lactobacillus* has been shown to increase the antimicrobial activity of Chinese chives, which is attributed to the production of metabolite concentration [11]. Like bacteria strains mentioned above, *Candida* and *Basidiomycota* also showed a positive association with key metabolites in fermented pine needles. These fungi catalyze the formation of esters. These substances then interact to produce the distinct fermented flavor of pine needles [42]. In conclusion, diverse metabolites produced in the fermented pine needles is attributed to the harmonious effect of bacteria and fungi.

In the present study, KEGG data illustrated that the ‘biosynthesis of phenylpropanoid, flavonoids, and secondary metabolites’ were the core functional pathways in fermented pine needles. Phenylpropanoid is the largest class of natural products, including flavonoids, anthocyanins, lignans, and tannins. It has a variety of functions, including photosynthesis, nutrient absorption, growth regulation, cell division, maintenance of redox homeostasis, and biological and abiotic stress reactions [43]. Flavonoids is a class of compounds abundantly present in plants kingdom and are responsible to regulate growth and development of plants in addition to other bioactivities with potential industrial implications [44]. Anthocyanins, chalcones aurones, and flavonols are major pigments (which endow plants with various colors) also belong to flavonoids class. Flavonoids possess a variety of uses in daily living, including food and medicine. Proanthocyanidins and anthocyanins, in particular, play an important role as edible pigments and flavor-modifying agents in wine and food [45]. Beyond their culinary applications, plant flavonoids have shown promise in slowing down the aging process in several body systems, including the skin, neurological system, reproductive system, liver, and immunological organs when employed as active ingredients. Furthermore, they are involved in the prevention of illnesses including Alzheimer’s, breast cancer, osteoporosis, and cardiovascular disorders [46]. Our findings indicate that differential metabolites observed in fermented pine needles are attributed to the interaction between the number of microorganisms increased by fermentation and fermenting microorganisms in pine needles at different fermentation stages.

5 Conclusions

The present study concluded that untargeted metabolomic analysis identified the differential metabolite contents of the fermented pine needles from LC-MS (430 metabolites) and GC-MS (278 metabolites) analyses. Major metabolites identified were associated with pathways including biosynthesis of phenylpropanoids, flavonoids, and secondary metabolites. Furthermore, the results demonstrated that major bacterial (*Firmicutes*, *Proteobacteria*, and *Lactobacillus*) and fungal (*Penicillium*, *Candida*, and *Basidiomycota*) taxa significantly correlated with the differential metabolites which might be attributed to the synergistic effect of both microbial taxa. Our findings provide practical insights into the mechanism of enrichment of metabolites in fermented pine needles and laid down the basis for developing traditional fermented foods with nutraceutical value.

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Availability of Data and Materials: Datasets used for this manuscript are available within the text and its supplementary files. The mass spectrometry proteomics data have been deposited to the ProteomeXchange Consortium (<http://proteomecentral.proteomexchange.org> (accessed on 14 October 2023)) via the iProX partner repository with the dataset identifier PXD047484.

Ethics Approval: Not applicable.

Conflicts of Interest: The authors declare that they have no conflicts of interest to report regarding the present study.

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