Nitric oxide alleviates cadmium-impeded growth by limiting ROS accumulation in pea seedlings

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Abstract: Cadmium (Cd) causes oxidative stress, which leads to the oxidation of various biomolecules by the production of reactive oxygen species (ROS) to facilitate programmed cell death (PCD). The antioxidant defense system fails to detoxify ROS when it is produced in excess. Nitric oxide (NO), a gaseous free radical and a phytohormone, regulates various physiological processes of plants. Therefore, this work was undertaken to study the effects of the application of exogenous sodium nitroprusside (SNP, a NO donor) on growth parameters, oxidative stress, accumulation of secondary metabolites, and activities of antioxidant enzymes under Cd stress. Mild (50 μ M) and severe (200 μ M) Cd stress were applied to hydroponically grown pea (Pisum sativum L.) plants with or without 50 µM SNP. Severe Cd stress had a substantial impact on the plants. The effectiveness of NO in reducing Cd-induced negative effects on plant height, fresh weight, dry weight, protein content, nitrite content, nitrate reductase (NR) activity, catalase activity, and peroxidase activity were investigated. Seedling development, protein content, nitrite content, nitrate reductase (NR) activity, antioxidant defense systems disruption, overproduction of reactive oxygen species, and oxidative damage were observed. The antioxidant defense system (catalase and peroxidase activities) was activated by NO, which resulted in lower lipid peroxidation and lower hydrogen peroxide (H2O2) levels in Cd-exposed plants. SNP treatment boosted endogenous NO levels and NR activity in Cd-stressed plants while also enhanced proline levels to preserve osmotic equilibrium. The presence of total phenols and flavonoids increased after SNP treatment, indicating that SNP enhanced stress recovery and boosted plant development in Cd-stressed plants.

Introduction

The toxic heavy metal like cadmium (Cd) reaches the soil mainly as Cd^{2+} via various natural and anthropogenic activities. Soil pollution occurs because Cd becomes a hazardous threat by inhibiting plant development by damaging various metabolic activities, reducing intake of water and nutrient, and modifying organ development. Furthermore, the availability of Cd in the soil causes a reduction in the market price of various edible plants, and

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as a result, Cd becomes a challenge for living beings (Zhang *et al.*, 2020). Cd simply reaches the root from the soil either through channel or nutrients transporters or via aquaporins, and from there, it is transported to aerial parts of the plants and gets accumulated. Cd has a great affinity to phosphates, porphyrins, purines, pteridines, histidyl, and cysteinyl side chains of proteins; hence, it has the potential to damage nucleic acids, proteins, lipids, and enzymes (Cheng *et al.*, 2016). Cd promotes severe stress by causing increased production of reactive oxygen species (ROS) (Kumar and Khan, 2021), which, if not detoxified, promptly lead to oxidation of proteins, lipids, and DNA, triggering programmed cell death (PCD), and dysregulation of a serious physiological and metabolic process of plant development leading to selective cell death.

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Plants usually trigger defense strategies at every stage of their life to protect themselves from toxicity in the soil; they protect themselves from heavy metal contamination by controlling the free radicle ion produced in the cell by eliminating, immobilizing, chelating, attaching to thiol-peptides, and by vacuolar segregation (Zaid and Wani, 2019). However, when ROS is produced in excess, the antioxidant enzymes fail to reduce their level. In heavy metal-stressed plants, attempts have been made to increase the efficacy of components of the antioxidant defense system by supplying various compounds and phytohormones exogenously (Mohamed et al., 2019; Sharma et al., 2012). In the past few years, several results have shown that the reactive nitrogen species (RNS), specifically nitric oxide (NO), interacts with ROS to protect the plant from heavy metals stress, although the mechanism behind this process is still unknown. In fact, while the ROS produce oxidative stress and NO may produce nitrosative stress, both act as signaling molecules depending on their amounts in the cells in response to Cd stimuli (Liu et al., 2019).

Plants can change their NO metabolism when exposed to Cd toxicity; nevertheless, conflicting data are available showing the effect of Cd^{2+} on endogenously produced NO (Genchi *et al.*, 2020). In fact, as reported in numerous plant species, Cd can either enhance or inhibit NO levels (Corpas and Barroso, 2014). Furthermore, exogenous NO donor treatments were demonstrated to defend plant tissues from oxidative stress that occurred due to Cd^{2+} through increasing ROS scavenging (Kopyra *et al.*, 2006; Noriega *et al.*, 2007).

NO, being a plant hormone has lately been recognized as a possible signaling molecule (Khan et al., 2017; Khan et al., 2022a; Khan et al., 2022b). It is a universal signal molecule that controls various activities of animals and plants at the physiological and molecular level and provides tolerance to plants from heavy metal stress by increasing the antioxidant defense system (Anjum et al., 2016; Nabi et al., 2019). Several findings have stated the involvement of NO in the regulation of plant Cd response (Kopyra et al., 2006). Despite the fact that NO is important for regulating the biological processes produced by plants against metalloid stress, the exact pathway of NO signaling is unknown to date. Sodium nitroprusside (SNP) is one among other extensively used NO donor substances for supplying NO to plants, regardless of whether they are growing in natural soil, synthetic or prepared media, or hydroponic (Soares et al., 2021). Its significance in physiological processes, however, is determined at the cellular level. Indeed, at very low levels, it serves as a signaling molecule, but at high levels, it acts as a stress-causing signaling molecule (Fancy et al., 2017).

The peroxynitrite (ONOO⁻) is formed when NO reacts with superoxide anion (O₂^{•-}), which is considered one of the simplest and fastest reactions in living systems (Arasimowicz-Jelonek and Floryszak-Wieczorek, 2011; Corpas and Barroso, 2014). The peroxynitrite is a highly unstable anion and a member of the RNS family. Although it is generally established that ONOO⁻ is not hazardous to plant cells at low concentrations, a rise in its cellular levels causes stress, as reported in Arabidopsis (Corpas and Barroso, 2014). NO has the ability to modify a variety of proteins, either directly through S-nitrosylation, nitration, and nitrosylation, or secondarily through monitoring the stress proteins genes transcription, which explains its role in plant physiological/metabolic processes (Fancy *et al.*, 2017).

Pea (*Pisum sativum* L.) is the main nutritious legume crop, considered highly sensitive to Cd toxicity compared to cereals and grasses. Based on earlier studies, we hypothesized that NO plays an important role in providing tolerance to pea plants from Cd stress. Therefore, we subjected pea seedlings to low and high Cd concentrations to understand the response and physiology of pea plants growing under Cd stress. Our primary objective was to use this understanding to produce Cd tolerance in other plant varieties that currently suffer from severe Cd damage and crop losses. We investigated the effect of Cd stress on pea seedlings and the mechanism by which exogenously supplied NO ameliorates Cd toxicity in examined plants.

Materials and Methods

Plant material and stress conditions

The sodium hypochlorite (NaOCl) (v/v) solution of 0.5% was used for surface sterilization of pea (Pisum sativum L.) seeds which were then rinsed thrice with deionized water. The sterilized seeds were placed at random on two-layered filter paper moistened with distilled water and kept at 28 ± 2°C for 3 days for better germination. After germination, seedlings were moved to pots having Hoagland solution. The treatment solutions included two concentrations of $CdCl_2$ (50 µM and 200 µM) with or without SNP (50 µM) added to the Hoagland solution. Growth media were continuously aerated and renewed on each third day. A growth chamber (ACMAS Technology Pvt., Ltd., India) was used for growing seedlings. The growth chamber was maintained with 12 h/12 h light/dark cycles, at 25/20°C, under a white fluorescent light intensity of 350 µmol photons m⁻² s⁻¹, and 70% relative humidity. After three weeks of growing, plants were harvested, washed completely, and plant parts were used for further estimations. All the experiment was carried out thrice.

Evaluation of growth and biomass

The shoots and roots length of pea seedlings were measured by a measuring scale. The length between root-shoot knot and the tip apex of the root was considered as root length, and shoot length was the difference between the base of the culm to the tip apex of the plants. The measurements were presented in centimeters. The seedlings were cleaned with double distilled water and blotted between filter papers before measuring their fresh weight (FW). A weighing balance was used to determine the fresh weight. Dry matter was obtained by drying the root and shoot samples at 70°C for 72 h in an oven.

Quantification of Nitric Oxide

By quantifying nitrite, NO was indirectly measured. Evans and Nason's method was used to calculate the nitrite content (Evans and Nason, 1953). Briefly, plant root and shoot were homogenized in acetic acid buffer (50 mM, pH 3.6) in a chilled mortar pestle, followed by centrifugation (10,000 g, 15 min) and removal of the residue. Greiss reagent was mixed with the obtained supernatant. This reagent contained 1.0% sulfanilamide in 1N-HCl, and 0.025% N-(1-Napthyl)-ethylene diammonium dichloride (NEDD) and optical density (OD) was recorded at 540 nm.

Determination of Nitrate Reductase (NR) activity

To determine NR activity, the method by Hageman and his colleagues was followed (Hageman *et al.*, 1980). Fresh root and shoot tissues (0.5 g) were crushed in potassium phosphate buffer (0.05 M, pH 7.8) and 0.4 M KNO₃ solution. The obtained homogenate was incubated at 35°C for 75 min in the dark. After incubation, tubes were subjected to a hot treatment for 5 min to terminate the activity of the enzyme. A 200 μ L aliquot of the resulted product was collected and mixed with 2 mL Greiss reagent (1.0% sulfanilamide in 1N HCl and 0.025% NEDD). Diazotization resulted in a pink color after 30 min of incubation. Finally, using distilled water, the volume was increased to 6 mL, and the OD was measured at 540 nm. The standard curve was made from sodium nitrite. The activity of the enzyme was measured in μ mol NO₂ min⁻¹ g⁻¹ FW.

Determination of Hydrogen Peroxide (H_2O_2)

The quantity of H_2O_2 was calculated using Cheeseman's approach (Cheeseman, 2006). In a prechilled pestle and mortar, 0.5 g of root and shoot was crushed in 5 mL trichloroacetic acid (TCA) buffer (0.1%) followed by centrifugation (12,000 g, 15 min, 4°C). The obtained supernatant was used to calculate the amount of H_2O_2 . One milliliter of supernatant was added to phosphate buffer (10 mM, pH 7.0), and KI (1 M), and the absorbance was measured at 390 nm. The concentration of H_2O_2 was measured in μ mol H_2O_2 g⁻¹ FW.

Estimation of Lipid Peroxidation

The method, using 2-thiobarbituric acid (TBA), developed by Heath and Packer (1968), was used for estimating the malondialdehyde (MDA) level, which is produced upon preoxidation of polyunsaturated fatty acid. Root and shoot were homogenized in 0.1% TCA buffer followed by centrifugation for 10 min. The supernatant obtained after centrifugation was incubated at 95°C with 4 mL of 0.5% TBA in 20% TCA for 30 min. To stop the process, the mixture was immediately placed on ice and centrifuged (8,000 g for 10 min). The MDA level was obtained by calculating the differences in absorbance at 600 nm and 532 nm using 155 mM⁻¹ cm⁻¹ as an extinction coefficient. The concentration of MDA was expressed as nmol MDA g⁻¹ FW.

Estimation of proline content

Root and shoot tissues (0.5 g) were homogenized in 10 mL of extraction buffer (3% sulfosalicylic acid). The resultant filtrate was used to calculate proline using the technique of Bates *et al.* (1973). In a test tube, 2 mL filtrate was added to 2 mL of acid ninhydrin followed by 2 mL of glacial acetic acid and incubated at 100°C for 1 h. The reaction was immediately transferred to the ice to stop the reaction. Toluene was used to extract the reaction mixture. A test tube stirrer was used to thoroughly mix the test tubes for 15–20 s. The toluene-containing chromophore was removed from the upper layer, and the OD was measured spectrophotometrically at 520 nm, considering toluene as a blank. L-proline (0.02 to 0.1 M)

standard curve was used to measure the concentration of proline and expressed as mg proline g^{-1} FW.

Estimation of protein content

Bradford's test was performed to estimate the total amount of protein in root and shoot samples (Prasertsongskun *et al.*, 2002). In test tubes, 100 μ L plant extract was mixed with 5 mL of the diluted dye. The reaction mixture was incubated at room temperature for 5 min, and the color change was recorded spectrophotometrically at 595 nm. The standard curve was prepared using bovine serum albumin to calculate the protein concentrations.

Estimation of antioxidant enzyme activities

The plant tissues were homogenized in phosphate buffer (100 mM, pH 7.5) with 1 mM EDTA and 5% insoluble PVP in a 3:1 ratio, followed by centrifugation at 10,000 g for 30 min. Protein and enzyme activities were measured using the extracts. The activity of catalase (EC: 1.11.1.6) was estimated by measuring the decline in absorbance that occurred due to the disappearance of H₂O₂ at 240 nm (Aebi, 1984). In a cuvette, 100 µL of enzyme extract was quickly added to 3.0 mL H₂O₂-phosphate buffer prepared by mixing 0.067 mM phosphate buffer (pH 7) with 2 mM H_2O_2 . The change in OD was measured in a spectrophotometer at 240 nm at intervals of 30 s up to 3 min. As a control, an enzyme solution containing H₂O₂-free phosphate buffer was used. The decreasing absorbance was measured at 240 nm for 3 min, and the enzyme unit mg^{-1} protein was calculated. One unit of the enzyme was defined as the quantity of enzyme essential to degrade 1 μ M H₂O₂ min⁻¹.

The activity of peroxidase (POD, EC: 1.11.1.7) was measured by using the procedure of Li *et al.* (2014). POD uses pyrogallol as a hydrogen donor and converts H_2O_2 to H_2O and O_2 . When pyrogallol is oxidized, it generates purpurogalli, a colorful product that may be detected spectrophotometrically at 430 nm. A hundred microliters of enzyme extract was mixed thoroughly with 3.0 mL of 50 μ M pyrogallol solution. Five hundred microliters of 1% H_2O_2 was mixed properly in the test cuvette. The change in OD at 430 nm was recorded for a gap of every 30 s till 3 min and presented as enzyme units mg⁻¹ protein. One enzyme unit was defined as the quantity of enzyme needed to produce one µmole purpurogalli every min.

Determination of phenolics and flavonoids

An assay modified by Shetty and co-workers was used to measure total soluble phenolics (Shetty *et al.*, 1995). The leaf tissue weighing 0.5 g was immersed in 95% ethanol and frozen for 48 h. The leaf sample was crushed and followed by centrifugation at 12,000 g for 10 min. The supernatant (0.5 mL) was diluted with 0.5 mL distilled water and then placed into a test tube having 1 mL 95% ethanol and 5 mL distilled water. Each sample was mixed thoroughly with 500 μ L of 50% (v/v) Folin and Ciocalteu's reagent. The sample mixture was incubated for 5 min, then 1 mL of 5% Na₂CO₃ was added to it and left for 60 min. As a blank, 0.5 mL of double distilled water was taken, and the absorbance was noted at 725 nm. The standard curves were prepared by taking various concentrations of gallic acid in 95% ethanol.

The obtained absorbances were transformed to total phenol and represented in mg equal to gallic acid per g of FW.

The total flavonoid content was determined using the method of Jia and coworkers (Jia *et al.*, 1999). At room temperature, the leaf sample was homogenized with ethanol. To each extract, methanol was added to 300 μ L to make the final volume up to 2 mL followed by the addition of 0.1 mL of AlCl₃ (10%) and 0.1 mL of 1 M sodium acetate (CH₃COONa). The final volume was then adjusted to 5 mL using double distilled water. The sample was then incubated for 30 min and absorbance was measured at 415 nm using a spectrophotometer (Perkin Elmer UV-VIS 35). The standard curve was made using quercetin. The data were given in μ g of quercetin per g of fresh weight.

Statistical analysis

One-way ANOVA was performed by Graph Pad PRISM version 5.01. The error bars on the figures represent the standard deviation (SE). Using Tukey's honest significant difference (HSD) test at $p \leq 0.05$, different letters show

significant differences while the same letters represent no significant differences.

Results

Growth and biomass yield

Cd concentration of 50 μ M decreased the shoot length by 37%, root length by 36%, shoot fresh weight by 21%, shoot dry weight by 26%, root fresh weight by 37%, and root dry weight by 21% in comparison to those of the control (Figs. 1A, 1C and 1E). Cd concentration of 200 μ M reduced the shoot length by 62%, root length by 47%, shoot fresh weight by 43%, shoot dry weight by 61%, root fresh weight by 60%, and root dry weight by 54% in comparison to those of the control. The use of 50 μ M SNP with 50 and 200 μ M Cd increased the shoot length by 46% and 77% and root length by 63% and 42%, respectively, compared to the treatment with 50 and 200 μ M Cd alone (Figs. 1A and B). The shoot fresh weight and dry weight were also elevated by the supplementation of 50 μ M SNP with 200 μ M Cd, and the



FIGURE 1. Effect of Cd, SNP, and their combinations on length (A,B), fresh weight (C,D), and dry weight (E,F) of shoot and root of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.

maximum increase in shoot fresh weight and dry weight was 42% and 65%, corresponding to 200 μ M Cd treatment (Figs. 1C–1F). The root fresh weight and dry weight also increased when 50 μ M SNP was used along with 200 μ M Cd, and an increase in root fresh weight by 56% and root dry weight by 64% was observed compared to those after 200 μ M Cd treatment (Figs. 1C–1F).

Nitrite and NR content

Cd concentrations of 50 and 200 µM reduced nitrite content by 53% and 60% in shoot and 50% and 68% in root compared to control. Supplementation of NO with Cd improved the nitrite content and NR activity in both root and shoot (Figs. 2A-2D). SNP concentration of 50 µM with 50 µM Cd enhanced the nitrite content by 23% and 27% in shoot and root, respectively, whereas 200 µM Cd enhanced the nitrite content by 24% and 59% in shoot and root, respectively (Figs. 2A and 2B). Cd treatment reduced the NR activity in shoot and root by 47% and 34%, respectively, at 50 µM concentration and by 100% and 61%, respectively, at 200 µM concentration compared to control. The use of NO to 50 µM Cd treated plants enhanced the NR content by 27% and 26% in shoot and root, respectively, whereas NO supplementation to 200 µM Cd stressed plants enhanced the NR content by 30% and 43% in shoot and root, respectively (Figs. 2C and 2D).

H_2O_2 and MDA content

In shoot and root, Cd treatment of 50 μ M led to a 31% and 44% increase in H₂O₂ and 16% and 44% increase in MDA content, respectively (Figs. 3A–3D). Cd concentration of 200 μ M caused 65% and 77% increase in H₂O₂ content and 44% and 120% increase in MDA content in the shoot and

root, respectively, compared to those in control plants. The addition of 50 μ M SNP to 50 μ M Cd reduced H₂O₂ accumulation by 24% in the shoot and 24% in the root, whereas the addition of 50 μ M SNP to 200 μ M Cd caused 28% reduction in H₂O₂ content in the shoot and 25% reduction in the root (Figs. 3A and 3B). MDA content decreased by 22% in shoot and 31% in root when 50 μ M SNP was used along with 50 μ M Cd compared to 50 μ M Cd treatment alone. Similarly, the use of 50 μ M SNP along with 200 μ M Cd reduced MDA content by 22% in the shoot and 22% in the root (Figs. 3C and 3D).

Proline and protein contents

The amount of proline reduced by 22% in the shoot and 28% in the root, whereas protein content decreased by 25% in the shoot and 39% in the root at 50 μ M Cd treatment (Figs. 4A–4D). At 200 μ M Cd concentration, proline content reduced by 5% in the shoot and 16% in the root, whereas protein content decreased by 63% in the shoot and 68% in the root, respectively, relative to those in control. Supplementation of 50 μ M SNP along with 50 μ M Cd led to a 100% and 42% decrease of proline content in shoot and root, respectively, and a 50% and 72% increase in protein content in shoot and root respectively. Treatment with 50 μ M SNP along with 200 μ M Cd decreased proline content by 73% in shoot and 31% in root and enhanced protein content by 74% in shoot and 100% in root compared to 200 μ M Cd treatment alone (Figs. 4A–4D).

Total phenols and flavonoids content

The contents of total phenols and flavonoids decreased by 41% and 37% in the shoot at 50 μM Cd and by 61% and 59% in the shoot at 200 μM Cd treated plants in



FIGURE 2. Effect of Cd, SNP and their combinations on nitrite content (A,B) and NR activity (C,D) in shoot and root of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.



FIGURE 3. Effect of Cd, SNP, and their combinations on H_2O_2 (A,B) and MDA (C,D) content in shoot and root of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.



FIGURE 4. Effect of Cd, SNP, and their combination on proline (A,B) and protein (C,D) content in shoot and root of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.

comparison to those in control plants (Figs. 5A and 5B). However, 50 μ M SNP increased the phenols and flavonoids by 30% and 20% in the shoot, respectively, when supplied along with 50 μ M Cd and by 40% and 36% in the shoot, respectively, when supplied with 200 μ M Cd (Figs. 5A and 5B).

Catalase (CAT) and peroxidase activities

In 50 μ M and 200 μ M Cd-treated stressed plants, CAT activity was reduced by 48% and 61%, respectively, in shoot and 34% and 62% in root as compared to control (Figs. 6A and 6B). POD activity was reduced by 10% and 19% in shoot and



FIGURE 5. Effect of Cd, SNP and their combination on phenols (A) and flavonoids (B) content in the shoot of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.



FIGURE 6. Effect of Cd, SNP, and their combination on catalase (A,B) and peroxidase (C,D) activity in shoot and root of pea seedlings. Different letters represent significant differences, and the same letters represent no significant difference by applying Tukey's honest significant difference (HSD) test at $p \le 0.05$.

root, respectively, in 50 µM Cd treated plants, but it was decreased by 48% and 36% in shoot and root, respectively, in 200 µM Cd treated plants (Figs. 6C and 6D). NO supplementation along with Cd resulted in a significant enhancement in the activities of CAT and POD both in root and shoot compared to 50 μM and 200 µM Cd treatment (Figs. 6A-6D). Supplementation of 50 µM SNP with 50 µM Cd increased CAT activity by 61% in shoot and 36% in root compared to 50 µM Cd treatment, whereas supplementation with 200 µM Cd led to a 76% increase in shoot and 58% increase in root compared to 200 µM Cd. The POD activity increased by 19% in shoot and 41% in root when 50 μ M SNP was supplied with 50 µM Cd and 34% in shoot and 44% in root when 50 µM SNP was supplied along with 200 µM Cd (Figs. 6C and 6D).

Discussion

The current findings suggest that NO has a role in regulating pea responses to Cd toxicity. The exogenous supply of NOdonor SNP to the Cd-contaminated rice root increases intracellular NO and alleviates the morphological and anatomical damages caused by Cd by restoring the ROS/RNS balance in the cell (Sharma *et al.*, 2020a). Plants absorb Cd easily through roots growing in Cd-rich environments. Only a little amount of Cd is transfered to the shoots from the roots (Moussa and El-Gamal, 2010). As Cd is a non-essential element, its accumulation inhibits the growth and development of the plant (Wang *et al.*, 2021). NO acts as a key signaling molecule in plants, and its importance in alleviating abiotic and biotic stressors has prompted widespread interest (Khan *et al.*, 2017). The endogenous NO level increased in Cd-affected pea plants developing in a stressed environment (Huang et al., 2020). This increase in NO suggests that it acts as a stress signaling molecule (Molassiotis et al., 2010). Treatment of Cd-stressed pea seedlings with SNP boosted endogenous NO levels even further, indicating effective uptake and storage of NO provided by SNP, which is consistent with earlier investigations that have found a similar role of SNP (Li et al., 2014). NO has been identified in recent studies as an important messenger molecule in plants and provides tolerance from various types of heavy metals like arsenic (Singh et al., 2009), Cd (Wang et al., 2013; Wang et al., 2013a), and copper (Zhang et al., 2009). In one previous study, we reported that a specific concentration of NO was needed to alleviate Cd stress in pea seedlings, and a low concentration of NO provides more tolerance against Cd toxicity (Khan et al., 2017; Mohamed et al., 2019). In a previous work, different concentrations of SNP were supplied exogenously to a Cd-treated growth medium to determine the optimal NO level for efficiently reducing Cd toxicity. Pea seedlings treated with a Cd cause a decrease in length and fresh and dry weight of root and shoot (Figs. 1A-1C). Cd toxicity reduces growth due to i) reduced absorption of water and nutrition by root (Wang et al., 2021), ii) damage to the photosynthetic membrane (Sharma et al., 2020b), iii), hindrance during cell division (Mondal et al., 2013), and iv) direct impedance of Cd with some hydrolytic enzymes, and changes in carbohydrates metabolism (Abdel Latef, 2013; Hussain et al., 2013). The exogenous use of SNP mitigates the inhibitory effect of Cd on length, and fresh and dry weight of root and shoot (Figs. 1A-1C), demonstrating that NO plays a direct or indirect function in reducing Cd toxicity in these plants. These findings are consistent with previous research on Typha latifolia and chickpea (Wang et al., 2016; Kumari et al., 2010).

The two probable enzymatic sources of NO synthesis are NR and NOS (Neill *et al.*, 2008). In plants, NR is regarded as the primary enzymatic source for the production of endogenous NO (Li *et al.*, 2016). Several studies have reported NOS activity in plants (Corpas and Barroso, 2013), despite the evidence showing that no NOS gene was reported in plants (Neill *et al.*, 2008). Endogenous NO levels and NR activity were reduced in Cd-treated pea seedlings (Figs. 2A and 2B). Cd causes a reduction in endogenous NO concentration and NR activity which was reversed after an exogenous supply of SNP (Figs. 2A and 2B).

Cd stress causes ROS overproduction leading to oxidative burst and damage to various biomolecules in plants. The outcome of lipid peroxidation is MDA, and its concentration reflects lipid peroxidation and stress level (Abdel Latef, 2011; Ahmad *et al.*, 2015). Enhanced MDA production and loss of membrane permeability have been reported in maize and pepper. In this study, we observed that Cd treatment caused the increase of H_2O_2 and MDA content in the root and shoot of pea (Figs. 3A and 3B). Similar to our studies, Cd has been reported to cause oxidative stress by enhancing H_2O_2 and MDA contents in *Arabidopsis* (Li *et al.*, 2016). H_2O_2 , on the one hand, act as a signaling molecule (Liu and He, 2016; Liu and He, 2017), on the other hand, it is very harmful to most biochemical reactions (Nahar *et al.*, 2021). H_2O_2 hinders the Calvin cycle by lowering the photosynthetic rate (Hussain *et al.*, 2013), and its high content is generally assumed as a stress condition in plants. When Cd-treated pea plants were supplied with 50 μ M SNP, plants showed a decrease in H_2O_2 and MDA levels than those in plants treated with Cd alone (Fig. 3A and 3B). This could be because of NO-led modulation in antioxidant enzymes activity and proline accumulation, which can detoxify free radicals and decrease oxidative damage of membranes during Cd stress (Mourato *et al.*, 2012; Li *et al.*, 2016).

Heavy metal exposure has been linked to a decrease in the level of soluble proteins (Hasanuzzaman et al., 2014). Our findings showed the reduction of protein content in Cd-treated plants. This may be because of (i) an increase in ROS generation, which damages proteins (Gajewska and Skłodowska, 2007), (ii) binding of Cd to the sulfhydryl group of protein damages the protein structure (Hasanuzzaman et al., 2014), and (iii) active involvement of protease activity (Palma et al., 2002). Plants treated with Cd in combination with 50 µM SNP displayed more protein content related to Cd-treated plants (Fig. 4B). Parallel to our studies, SNP-mediated increase in protein content has been observed under Cd stress in peanuts (Dong et al., 2019) and seawater stress in canola plants (Abdel Latef, 2011). SNP improved the content of protein during abiotic stress, perhaps by increasing protein synthesis and lowering proteolysis and enzyme degradation (Kozlowski and Pallardy, 2002).

Phenolic compounds and flavonoids are well known for their protective role against various stresses in plants. Total phenols and flavonoids represent antioxidant properties because of their capacity to act as electron-donating agents (Salinitro et al., 2020). Flavonoids have also been implicated in metal chelation (Bai et al., 2004). With an increase in Cd concentration, phenols and flavonoid levels decreased in the roots and shoots of pea seedlings (Figs. 5A and 5B). Similar to our studies, the reduction of phenols and flavonoid levels due to Cd was also reported earlier (Kapoor et al., 2014). Cd toxicity causes the overproduction of ROS by inhibiting the enzymes involved in the production of phenols and flavonoid. Phenolic compounds are produced immediately through the signaling processes during stress (Bais et al., 2002). Exogenous supply of SNP along with Cd enhanced the total phenol and flavonoid level in roots and shoots of pea seedlings (Figs. 5A and 5B). This may be probably due to SNP-led (i) decrease in Cd uptake, (ii) enhancement in polyphenols biosynthesis genes expression (Xu et al., 2014), and (iii) stimulation of the activity of phenylalanine ammonia-lyase enzyme, a key enzyme in phenyl propanoid biosynthesis (Kuthanová et al., 2004).

Antioxidant enzymes, including CAT and POD play important roles in plants by avoiding oxidative damage (Lu *et al.*, 2020). Both CAT and POD, decompose H_2O_2 to H_2O and O_2 (Ahmad *et al.*, 2015). Compared to CAT, POD possesses a high affinity for H_2O_2 . CAT is responsible for the elimination of excess H_2O_2 , and POD is responsible for keeping H_2O_2 concentrations low (Mourato *et al.*, 2012; Hasanuzzaman *et al.*, 2020). Our results showed decreased CAT and POD activity in Cd stress pea seedlings (Figs. 6A and 6B). Excessive production of ROS during Cd stress may be the cause of the decrease in antioxidant enzyme activity (Mondal *et al.*, 2013). The use of SNP increased antioxidant enzyme activity and lowered H_2O_2 and MDA content suggesting that SNP application can be effective in increasing Cd stress tolerance in the plant by enhancing antioxidant systems and protecting the structural and functional damage of cell membranes (Talukdar, 2012; Piacentini *et al.*, 2020).

Conclusion

Our results suggest that the application of SNP along with Cd increases endogenous NO content and antioxidant activity, which might be responsible for the mitigation of Cd-impeded growth in pea seedlings.

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