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ARTICLE

Photodegradation and Removal of Diclofenac by the Green Alga Nannochloropsis oculata

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ABSTRACT

The emerging contaminants (ECs) are organic compounds including pharmaceuticals. These products are consumed in large quantities, favoring their continuous entrance to the environment and affecting the health of living beings. Diclofenac is a non-steroidal anti-inflammatory drug with analgesic properties of prolonged release. It is the commonest postoperative analgesic, and it is highly consumed without prescription. In recent years, the capability of microalgae to remove pharmaceuticals in water has been tested as a promising tool. In this work, the removal efficiency of diclofenac (16 µg/mL) by the microalga *Nannochloropsis oculata* CCAP 186/7 was evaluated. The major nutrient concentrations in Bold's Basal Medium were modified (nitrogen and carbon: 50 and 100% of their original concentration) to know their effect in the removal of diclofenac. Drug degradation by light was also evaluated. The maximum removal capacity of diclofenac in the medium was 68.8%. The percentage of diclofenac adsorbed on the cell wall was between 6 and 12.7%. Around 23.6% of diclofenac was degraded by light after 18 days. *N. oculata* did not show growth inhibition due to the presence of diclofenac. The results obtained point to the promissory application of *N. oculata* as a bioremediation system for diclofenac removal.

KEYWORDS

Adsorption; emerging contaminants; microalgae; pharmaceuticals; phycoremediation; photodegradation

1 Introduction

Since the early eighties, many studies have shown that emerging contaminants (ECs) and their transformation products are present in aquatic environments worldwide [1–4]. ECs include pharmaceuticals, personal care products and pesticides, among others [5–6]. Within ECs, the pharmaceutically active compounds (PhACs) are of major concern. Although their environmental concentrations are usually very low, this kind of ECs represent a risk to both human health and the environment since they are not completely assimilated by the organism and are excreted with urine and fecal matter either as intact substances and metabolites.

The non-steroidal anti-inflammatory drugs (NSAIDs) have been found at concentrations of 5 ng/L to 4.1 μ g/L in samples of wastewater (WW), sewage sludge from urban wastewater treatment plants (WWTPs), surface water, groundwater, and drinking water [7,8]. One of the most widely used NSAIDs is



diclofenac, which has anti-inflammatory, analgesic, and antipyretic properties; it is fast absorbed by the duodenum, with a plasma half-life that oscillates around 2 h, and it is over the counter [9,10].

Diclofenac has negative effects such as impaired kidney function and necrosis in gill cells; it has been found accumulated in liver, kidneys, gills, and muscle tissue of fish [11,12]. The death of vultures of the genus Gyps associated with kidney failure in South Asia and Africa has been reported due to the ingestion of remains contaminated with diclofenac; the drug was detected in concentrations ranging from 0.051 to 0.643 μ g/L in kidneys of these vultures [13].

Due to the deficiency in conventional WWTPs to remove diclofenac, alternative strategies need to be implemented to improve water quality; since it returns to the environment, it can be incorporated into the food chain, generating toxic effects even at low environmental concentrations [14,15]. A viable alternative for NSAIDs removal and other drugs is the phycoremediation, which comprise the use of macroalgae, microalgae and cyanobacteria for the removal of nutrients and xenobiotics from wastewater and carbon dioxide from the air [16]. Microalgae-mediated bioremediation of ECs has gained attention in recent years, due to its ability to convert CO_2 into biomass assimilating large amounts of inorganic nutrients (e.g., phosphorus and nitrogen). Furthermore, relatively small amounts of operational inputs are required. Therefore, the use of algae is a sustainable and environmentally friendly technology [17–19].

Microalgae have a great flexibility to grow and survive under extreme environments; their metabolism can change between autotrophic and heterotrophic depending on the carbon sources and nutrients in the environment; it has been proposed that this adaptability makes them great candidates for removal of ECs [20–22].

The mechanisms for ECs removal by microalgae include bioaccumulation, bioadsorption, and intracellular and extracellular biodegradation. Moreover, photolysis plays an important role in the elimination process of some pharmaceuticals [14,23,24]. Several microalgae species have been employed for diclofenac removal, including *Chlorella sorokiniana, Chlorella vulgaris, Chlorella pyrenoidosa, Chlamydomonas reinhardtii*, and *Scenedesmus obliquus*. These species have been evaluated in axenic cultures in different growing conditions; diclofenac removal percentages achieved by these microalgae were up to 79% in batch experiments [3,25,26].

Nannochloropsis is one of the commonest green algae found in different types of water systems; this genus has high photosynthetic efficiency and growth rate, good environmental adaptability, contamination resistance, and it is able to growth mixotrophically [27]. In recent years, some studies have been carried out with *Nannochloropsis* sp. in bioremediation processes for the elimination of PhACs (i.e., trimethoprim, sulfamethoxazole, carbamazepine, ciprofloxacin, and triclosan) in water, obtaining removal percentages from 60 to 100% [28,29]. Yoshida et al. [30] demonstrated the capacity of formaldehyde removal by the strain *N. oculata* ST-3, which was able to degrade 99.3% of formaldehyde from the medium at a concentration of 19.9 μ g/mL. It has been proposed that the diclofenac can be taken up by microalgae as a carbon source [25,31,32].

The present study focuses on the evaluation of diclofenac removal from the growth media, considering adsorption of the drug to the cell wall and photodegradation. Diclofenac removal was evaluated modifying the initial concentrations of two macronutrients (C and N applied each to 50 or 100% of their original composition) to know if their deficit generates greater drug removal. Therefore, the objective of this work was to determine the ability of *N. oculata* to remove diclofenac from the modified culture medium.

2 Materials and Methods

2.1 Microalgae Cultivation and Inoculum Preparation

The freshwater microalga *Nannochloropsis oculata* CCAP 846/7 was obtained from the Autonomous University of Aguascalientes, México. To maintain the stock culture, the microalga was inoculated into 100 mL of the Bold's Basal Medium (BBM) in sterile conditions at a volume ratio of 10%

 $(V_{\text{inoculum}}/V_{\text{medium}})$. The flask was stirred at 80 rpm and kept at 25°C for two weeks under a photoperiod of 16 h light and 8 h dark. The culture was grown under fluorescent lamps with an intensity of 150 µmol of photons per m² per second.

The microalga was cultivated until the exponential phase was reached and the absorbance adjusted at 0.2 AU at a wavelength of 750 nm using a GloMax[®]-MultiMicroplate Reader (PromegaCorporation, Madison, WI, USA) to generate the inoculum for further experiments.

2.2 Experimental Design

Batch experiments were performed by modifying the initial concentration of two nutrients to 50 and 100%. These nutrients were nitrogen (as NaNO₃, the main source of nitrogen) of Bold's Basal medium (BBM, composition in Tab. S1 of the Supplementary Material) and carbon. As the BBM does not contain a carbon source, the inorganic carbon source (CO_2) was supplied by aeration. The inoculum used was described in Section 2.1. The experiments carried out were: A) N. oculata grown with 50% of nitrogen and 100% of carbon (N50 C100), B) N. oculata grown with 100% of nitrogen and 50% of carbon (N100 C50), and C) N. oculata grown with complete medium (100% of nitrogen and 100% of carbon, N100 C100); each condition was run in triplicate. A standard solution of diclofenac sodium salt (Sigma–Aldrich, St. Louis, MO, USA) was prepared at a concentration of 250 µg/mL, diluted in 3 mL of ethanol, and supplemented with distilled water to obtain the desired concentration. The necessary volume of the diclofenac standard solution was added to the flasks to reach the initial concentration of 16 μ g/mL. The flasks were stirred at 80 rpm and kept at 25°C under a photoperiod of 16 h light and 8 h dark. The culture was grown under fluorescent lamps with an intensity of 150 μ mol of photons per m² per second. A control for each experiment was carried out under the same conditions but without diclofenac. These experiments were performed in 125 mL Erlenmeyer flasks containing 70 mL of culture medium that were inoculated with 10% of a microalgal cell suspension ($V_{\text{inoculum}}/V_{\text{medium}}$) and closed with sterile cotton-wool stoppers. The carbon supply, agitation, and aeration were kept constant by an air diffusion system designed with plastic hoses connected to an air pump set to 2.2 L/min. The uniformity of the airflow was controlled by flow-regulating valves of 0.25 in.

From each Erlenmeyer flask, a sample of 2.5 mL was withdrawn every 48 h to determine cell concentration of *N. oculata*, chlorophyll a concentration, and diclofenac concentration using HPLC analysis. All batches were run until the algal growth curves reached the stationary phase.

The diclofenac concentration was previously established by a phycotoxicity test where different concentrations of diclofenac were evaluated independently for N. *oculata* CCAP 846/7 in BBM media, selecting the concentration that induced the greatest growth (data not shown).

2.3 Determination of Cell Growth and Chlorophyll a

The *N. oculata* CCAP 846/7 growth was monitored by counting cell density in a Neubauer chamber. The specific growth rate (μ) was calculated as described by Zhu et al. [33].

Chlorophyll a was measured by a previously established methodology with some modifications [34]. Briefly, 1 mL of culture was harvested by centrifugation at 13,000 rpm for 5 min. The supernatant was discarded, and the pellet was re-suspended in 1 mL of methanol, sonicated at 50% amplitude for 60 s (Ultrasonic Processor GEX 130, CT, USA), and incubated at 4°C for 24 h under dark conditions; afterwards, the samples were centrifuged again for 5 min at 13,000 rpm. The absorbance of the supernatant was registered at 652, 665, and 750 nm using a UV-visible spectrophotometer (JenwayTM 6705, Cole-Palmer, OSA, UK). Chlorophyll a concentration of each sample was calculated by the equation described by Porra et al. [35].

2.4 Diclofenac Removal by N. oculata

To measure diclofenac removal in the culture medium, 1 mL of the microalgal samples withdrawn every 48 h was centrifuged at 13,000 rpm for 5 min; the supernatant was transferred to a new tube and evaporated (Vacufuge Plus, Eppendorf, HH, DE) at 30°C for 3 h. Afterwards, the pellet obtained was resuspended in 500 μ L of 100% ethanol and sonicated at 50% amplitude for 15 s. Following centrifugation (13,000 rpm for 5 min), the supernatant was removed, completely evaporated, and stored at room temperature until further use.

The samples were resuspended in 500 μ L of 100% ethanol and analyzed in a High-Resolution Liquid Chromatography (HPLC) equipment (Agilent 1260 Infinity) coupled to a diode array detector (DAD) (Santa Clara, CA, USA).

A column packed with hydrophobic spherical silica (Agilent, Eclipse XDB C-18) was used. A volume of 20 μ L of sample or standard was injected into the column fed with a 0.1% trifluoroacetic acid (TFA) aqueous solution (mobile phase A). An acetonitrile gradient was generated using 0.1% TFA in acetonitrile (mobile phase B) from 0 to 100% in a period of 10 min. A calibration curve was elaborated at diclofenac (Sigma–Aldrich, St. Louis, MO, USA) concentrations of 1, 2, 5, 10, and 20 μ g/mL.

2.5 Bioadsorption of Diclofenac

In order to evaluate the amount of diclofenac adsorbed on the cell wall of *N. oculata* CCAP 846/7, 1 mL of culture was used. The sample was centrifuged at 13,000 rpm for 5 min, the microalgae pellet obtained was dried at 70°C for 2 h and the dry weight registered. Subsequently 200 μ L of 1 M NaCl was added and the sample vortexed for 5 min. The sample was centrifuged at 13,000 rpm for 5 min and the supernatant recovered. The microalgae pellet was dried again and after recording the dry weight, the washing process was repeated with 200 μ L of 0.1% Tween-20 solution.

The supernatants recovered were concentrated in an evaporator for 3 h at 30°C. The resulting pellet was resuspended in 500 μ L of 100% ethanol and sonicated for 15 s at a 50% amplitude; subsequently, the sample was centrifuged at 13,000 rpm for 5 min and the supernatant recovered for a final evaporation for 24 h at 70°C. Once evaporated, the sample was resuspended in 500 μ L of ethanol and analyzed by HPLC as described above. This procedure was standardized for *N. oculata* with the addition of diclofenac.

The biomass was washed with 1 M NaCl and 0.1% Tween-20 to discriminate possible electrostatic and hydrophobic interactions between diclofenac and cell wall components. The dry weight was recorded to homogenize the 3 cultures and compare the results obtained from the concentration of adsorbed diclofenac.

2.6 Photodegradation Test

The abiotic removal of diclofenac was monitored in an Erlenmeyer flask with $16 \mu g/mL$ of diclofenac in 70 mL of BBM, but without microalgal inoculum. A negative control was performed, without microalgal inoculum and without exposure to light. The cultures were kept at the same conditions of incubation mentioned above. Samples were withdrawn every 48 h and the concentration of diclofenac in the culture medium was determined by HPLC. The experiment was carried out by triplicate.

2.7 Statistical Analysis

All experiments were carried out by triplicate with three biological replicates. The data obtained were analyzed by one-way analysis of variance (ANOVA) using Dunnett tests in the Minitab v.16 software. Values were considered significant for p < 0.05.

3 Results and Discussion

3.1 Biomass Growth

Fig. 1 shows the number of cells per mL of *N. oculata* CCAP 846/7 cultures exposed to an initial diclofenac concentration of 16 µg/mL and under different initial concentrations of nitrogen and carbon. Previous studies in our laboratory indicated that this concentration is not toxic for *N. oculata* CCAP 846/7; even more, the growth of this microalga was stimulated. As it can be observed in Fig. 1, all

cultures exhibited the same growth behavior. In the case of the N50 C100 culture, lower growth was observed in both the control and diclofenac cultures (Fig. 1a). However, there was no significant difference between the control and diclofenac cultures in this treatment (Fig. 1a). After 14 days of growth in the N100 C50 culture, the number of cells/mL was significantly greater in the diclofenac than in the control culture (Fig. 1b). Also, there were no significant difference between the control and diclofenac cultures in the N100 C100 treatment (Fig. 1c). The slight decreases in growth in the control in the cultures with N100 C50 and N100 C100 (Figs. 1b and 1c) may be explained by the fact that the microalga can use the diclofenac as an additional source of organic carbon it is well known that the Nannochloropsis genus have a mixotrophic growth. It has been previously reported that the addition of some pharmaceuticals stimulated microalgae growth during batch cultures [2,25]. However, it has been reported that CO₂ metabolism is strongly related to nitrogen assimilation. For this reason, since nitrogen is the limiting nutrient; it could affect the consumption of the carbon source. Due to this, it was inferred that a part of the diclofenac removed in the N50 C100 culture could be internalized and stored in vacuoles, instead of being metabolized [36]. After 16 days, the stationary phase was observed in the cultures exposed to diclofenac and in the control; optical density values (Fig. S1 in Supplementary Material) and the number of cells per milliliter remained constant.



Figure 1: Growth kinetics displaying changes in number of cells per mL for *N. oculata* exposed to diclofenac (16 µg/mL) over time under different concentrations of nitrogen and carbon. The control cultures did not contain diclofenac. a) 50% of nitrogen and 100% of carbon (N50 C100), b) 100% of nitrogen and 50% of carbon (N100 C50), and c) 100% of nitrogen and 100% of carbon (N100 C100). Error bars indicate standard deviations. Statistically significant differences from the control are indicated by an asterisk (ANOVA, p < 0.05, Dunnett test)

The specific growth rate (μ) for *N. oculata* CCAP 846/7 growing with and without diclofenac was calculated (Tab. 1); according to the results, no significant differences (p > 0.05) were detected between the controls and their corresponding treatment with diclofenac N50 C100 and N100 C50. The highest μ was reached in the treatment having N100 C100 with diclofenac.

Table 1: Specific growth rate μ [d⁻¹] for *Nannochloropsis oculata* CCAP 846/7 under different concentrations of nitrogen and carbon. The control cultures did not contain diclofenac. Results are given as average value ± standard deviation. An asterisk indicates a statistically significant difference (ANOVA, *p* < 0.05, Tukey test)

Treatment	Control [d ⁻¹]	Diclofenac [d ⁻¹]
N50 C100	0.116 ± 0.007	0.114 ± 0.009
N100 C50	0.095 ± 0.003	0.102 ± 0.035
N100 C100	0.103 ± 0.008	$0.120\pm0.001\texttt{*}$

Note: (Concentration: 100% N = 250 mg/L; 100% C = 0.33 mg/L).

The concentration of chlorophyll a was quantified in cultures exposed to diclofenac and controls under different nutrient conditions. As it can be seen in Fig. 2, all cultures showed accumulation of chlorophyll a during the exponential growth phase; the concentration of this pigment gradually decreased between days 8 and 14, where the beginning of the stationary phase was appreciated. These differences were attributed to the nutrient deficit, since the N100 C100 culture showed greater accumulation of chlorophyll a. According to Fig. 2, there were no significant differences in the chlorophyll a concentration in the control cultures and those with 16 μ g/ mL of diclofenac; this suggests that the presence of this drug did not cause a negative effect in the pigment concentration during the growth of *N. oculata* CCAP 846/7. These data are in accordance with those reported by de Wilt et al. [3]; their studies using *Chlorella sorokiniana* indicated that the administration of a mixture of drugs (~100–350 μ g/L), including diclofenac, did not produce changes in the chlorophyll a concentration. Also, Schmitt et al. [4] evaluated the phytotoxicity of diclofenac at a concentration of 50 μ g/mL in *Scenedesmus vacuolatus*, without finding any modification in the concentration of chlorophyll a. Although it is a different genus, they showed a similar behavior.

Regarding the nutrient concentration, the culture N100 C100 (Fig. 2c) showed the highest concentration of chlorophyll a, meanwhile, the culture N100 C50 (Fig. 2b) presented the lowest concentration. This suggests that changes are occurring in the photosynthetic apparatus to adapt to the carbon deficient medium. Some authors have mentioned that carbon deficiency can cause changes in the photosynthetic apparatus, in the synthesis of carbohydrates, and in the way of capturing and fixing carbon and oxygen present in the atmosphere. In the absence of this compound there is practically no growth, or there is a limitation in productivity when the supply of this compound is deficient [37,38]. These results are not consistent with those reported by Huertas et al. [39], who evaluated the growth and accumulation of chlorophyll a in cultures of *Nannochloropsis gaditana* and *N. maculata* aerated with different concentrations of CO₂. The chlorophyll a concentration group (1% v/v CO₂). However, cultures of *N. maculata* with low carbon concentration (<0.0001% v/v CO₂) showed an increased chlorophyll a accumulation. Also has been reported that CO₂-enriched aeration promotes growth of *Nannochloropsis sp.*, obtaining a higher biomass yield in aerated cultures with a higher percentage of CO₂ compared to aerated cultures with ambient CO₂ levels (~0.03%) [40].

On the other hand, the culture N50 C100 (Fig. 2a) showed a decrease in chlorophyll a concentration with respect to the culture N100 C100 (Fig. 2c). These results corroborate those reported by Msanne et al. [41], who determined that cultures of *C. reinhardtii* deprived of nitrogen have a decrease in the concentration of chlorophyll. Furthermore, studies in *Nannochloropsis sp.* and *Nannochloropsis oculata* indicate that nitrogen

restriction produces a decline in chlorophyll a concentration [42]. According to Pérez et al. [43], nitrogen is the nutrient that determines growth of microalgae since it is necessary for the fixation of CO_2 in autotrophic cultures or for the assimilation of carbon in heterotrophic cultures. However, this situation does not agree with what obtained in this work where no significant differences were found between the culture with deficiency of N or C (Figs. 2a, 2b); therefore, it can be inferred that the nutrients were still in an adequate proportion to meet the requirements of this microalga.



Figure 2: Chlorophyll a concentration in cultures of *N. oculata* CCAP 846/7 exposed to diclofenac (16 μ g/mL) under different concentrations of nitrogen and carbon. The control culture did not contain diclofenac. a) N50 C100, b) N100 C50, and c) N100 C100. Error bars represent standard deviations. Statistically significant differences from the control are indicated by an asterisk (ANOVA, *p* < 0.05, Dunnett test)

When comparing cell density and optical density graphs with the chlorophyll a graphs, a different behavior was observed. The chlorophyll a graphs showed less growth or started their phase of decline days earlier than those corresponding to the cell and optical density graphs. This has been reported by Bolier et al. [44] in studies with microalgae where the optical density is used as a measure of growth; the OD_{750} values may continue to increase even though the chlorophyll concentration decreased. The relationship between cell density and the information provided by other methods, such as chlorophyll concentration, is not constant. In the case of chlorophyll a concentration it depended on the physiological state of the cells since it provided a measurement of the amount of active microalgae biomass. However,

in the measurement of cell density by determining OD, some factors such as dead cells may interfere [45]. The optical density measurement is considered an indirect measurement; it can be affected by some type of contamination or dead microalgae cells, while the chlorophyll a (being the main photosynthetic pigment) is a reliable and commonly used indicator to estimate the total alive biomass of cultures.

3.2 Diclofenac Removal by N. oculata CCAP 846/7

In this study, the diclofenac removal was monitored in each batch experiment every two days; this concentration decreased over time in all treatments (Fig. 3). In the three experiments performed, the removal curves displayed a similar behavior.



Figure 3: Percentage of diclofenac remaining in the medium during cultivation time by *N. oculata* CCAP 846/7. Error bars indicate standard deviations. Statistically significant differences from the control are indicated by an asterisk (ANOVA, p < 0.05, Tukey test)

The highest removal of diclofenac was observed during the first 10 days of the kinetics; afterwards, the diclofenac concentration in the medium remained relatively constant (Fig. 3). In the culture N50 C100 we observed a higher removal of the drug during all the kinetic; however, significant differences were only found in the percentage of diclofenac in the medium on days 2 and 4 when considering all experiments.

The percentage of diclofenac removal by *N. oculata* CCAP 846/7 obtained after 10 days ranged between 49 and 59%, while after 18 days of cultivation the final percentages of removal were between 66 and 69%. These results are similar to those reported by Escapa et al. [25], who evaluated the removal of diclofenac by three strains of microalgae in batch cultures using a photobioreactor with 250 mL of Mann and Myers medium and an initial diclofenac concentration of 25 μ g/mL. At the end of the batch culture (17 days), the treatments with *Chlorella sorokiniana, Chlorella vulgaris*, and *Scenedesmus obliquus* showed removal efficiencies of 65, 69, and 98%, respectively. The authors suggested that these differences may be due to specific characteristics of the strains such as cell size.

The ability to remove a mixture of drugs (including diclofenac) by *C. sorokiniana* was evaluated by de Wilt et al. [3]. The microalga strain was cultivated in urine (23 days), water treated by anaerobic processes, and synthetic urine (31 days); these cultures were spiked with a mixture of drugs (diclofenac, ibuprofen, fenoprofen, trimethoprim, paracetamol, metoprolol, diaveridine, carbamazepine dihydrocarbamazepine, trimethoprim-d9, estrone, ethinylestradiol 17 β -estradiol, and 17 β -estradiol-d3) in a concentration range from 100 to 350 µg/L. At the end of the batch culture, the diclofenac removal ranged between 40 and

60%. However, similar removal percentages were observed in abiotic cultures. The authors suggested that the elimination of diclofenac could be attributed mainly to phototransformation [3].

3.3 Bioadsorption of Diclofenac by N. oculata CCAP 846/7

The bioadsorption of diclofenac by the cell wall of microalgae has been reported. Algae cell walls have a set of polymers with similarity to cellulose, pectin, hemicellulose, and lignin [23]. In this study, the concentration of diclofenac adsorbed by *N. oculata* was quantified every 48 h from cultures exposed to the drug (16 μ g/mL) under different concentrations of nitrogen and carbon. Over the time, the percentages desorbed with 0.1% Tween-20 were higher than those using 1 M NaCl; additionally, during the first days of growth (days 0 to 4) the concentration of diclofenac adsorbed was higher than that obtained afterwards (Tab. 2). This could be due to the relationship between the concentration of the drug available in the medium. Over time, the concentration of *N. oculata* cells increased and that of diclofenac decreased, resulting in less adsorption of the drug. This was more evident in the experiments N50 C100 and N100 C50; in the experiment N100 C100 the amount of diclofenac desorbed remained almost constant throughout the time.

Table 2: Percentages of diclofenac desorption from *Nannochloropsis oculata* CCAP 846/7 biomass under different concentrations of nitrogen (N) and carbon (C) over time. Values with different letters indicate statistically significant differences in the percentage of removal between solutions for each experiment (ANOVA, p < 0.05, Tukey test)

	% desorbed with 1 M NaCl		% desorbed with 0.1% Tween-20			
Time (Days)	N50 C100	N100 C50	N100 C100	N50 C100	N100 C50	N100 C100
0	5.34 ± 0.59^a	5.34 ± 0.48^a	3.85 ± 0.44^a	5.09 ± 0.58^a	6.31 ± 0.30^a	4.95 ± 0.12^a
2	4.87 ± 0.39^{a}	5.04 ± 0.32^{a}	4.01 ± 0.35^a	6.03 ± 0.62^a	$7.58\pm0.28^{\rm a}$	4.80 ± 0.66^a
4	4.23 ± 0.28^{a}	4.59 ± 0.20^{b}	3.21 ± 0.37^a	5.71 ± 0.50^{a}	$8.08\pm0.52^{\rm a}$	4.27 ± 0.24^{a}
6	3.97 ± 0.01^{a}	3.73 ± 0.35^a	4.05 ± 0.27^a	6.34 ± 0.63^{b}	6.58 ± 0.33^{b}	4.07 ± 0.24^{b}
8	2.83 ± 0.40^a	2.96 ± 0.13^a	3.84 ± 0.10^a	4.63 ± 0.08^{b}	5.22 ± 0.36^{b}	3.66 ± 0.21^a
10	3.19 ± 0.28^a	2.93 ± 0.38^a	3.28 ± 0.30^a	4.05 ± 0.80^{b}	6.20 ± 0.58^{b}	3.29 ± 0.65^a
12	3.23 ± 0.01^a	3.70 ± 0.40^a	3.61 ± 0.25^a	4.70 ± 0.40^a	5.73 ± 0.58^{b}	4.53 ± 0.44^a
14	3.32 ± 0.45^a	2.51 ± 0.38^a	3.88 ± 0.41^a	4.93 ± 0.37^a	8.44 ± 0.43^{b}	4.46 ± 0.25^a
16	2.55 ± 0.60^a	2.99 ± 0.34^a	2.91 ± 0.41^a	5.15 ± 0.47^{b}	5.01 ± 0.15^{b}	4.72 ± 0.48^{b}
18	2.41 ± 0.57^{b}	2.59 ± 0.38^{b}	2.33 ± 0.16^{b}	5.52 ± 0.41^{a}	6.68 ± 0.67^{a}	4.56 ± 0.44^{a}

In general, the percentage of drug retained by the cell wall in relation to the initial concentration was from 6.6-12.7%; these range was obtained after washing the biomass with both solutions. These results are in agreement with those reported by de Wilt et al. [3] who evaluated the removal of a mixture of drugs by *C. sorokiniana*. Regarding diclofenac, the percentage adsorbed in relation to the initial concentration was 7.5 and 5.5% in urine and in anaerobically treated black water, respectively.

The adsorption process varies significantly according to the hydrophobicity, structure, and functional groups present in the contaminating compounds and on the cell surface of the microalgae. Diclofenac is a weak acid with a pKa value of 4.7 [46]. Therefore, it was negatively charged during the experiments (the initial pH of the cultures was 6.8, while it slightly diminished at the end to 6.4–6.5). Protonated diclofenac shows low water solubility [47]; this implies that it also has a hydrophobic region. The cell

wall of algae has a complex constitution; specifically, in the case of *N. oculata* the cell wall contains highly aliphatic biopolymers called algaenans [48]. Moreover, this cell wall could contain amino acids, which were reported for *N. gaditana* [49]. Therefore, diclofenac could interact with the cell wall either electrostatically (by interaction with positively charged amines from basic amino acids) or hydrophobically (by interaction with algaenan).

The low percentage of diclofenac in the cell wall of *N. oculata* with respect to the percentage of drug removed from the culture medium suggests that other mechanisms could be participating in the remotion, including an intracellular biodegradation during which *N. oculata* could internalize and metabolize diclofenac in order to use it as a carbon source. There are not studies that analyze the interaction of microalgae with diclofenac on their cell wall.

3.4 Photodegradation of Diclofenac

It has been reported that diclofenac is among the drugs most susceptible to direct photolysis [24,50–51]. In this work, the removal of diclofenac via abiotic factors was also evaluated by supplying this drug into the medium without any microalgal inoculation. In the negative controls, no significant difference was found between the initial and final concentration of diclofenac. Therefore, it is inferred that drug degradation was associated to light. Fig. 4 shows the percentage of diclofenac degradation by light, which displays a decrease over the time. During the first 4 days, the drug concentration in the culture medium remained constant; however, from day 6 the concentration of diclofenac showed a continuous decrease. At day 12, 13.7% of the drug was degraded. The total percentage of diclofenac degraded was approximately 23.6% after 18 days.



Figure 4: Photodegradation of diclofenac in an abiotic culture exposed to a photoperiod of 16 h in light and 8 h in dark for 18 days

However, this contrasts with what was reported by Ben-Ouada et al. [1]. These authors evaluated the elimination of diclofenac under similar incubation conditions to this study (they used initial concentrations of 25, 50, and 100 μ g/mL) by two isolated microalgal strains, while the controls (without algae cells) served also for the determination of the diclofenac abiotic removal. Their results indicate that the abiotic removal did not exceed 8% after 5 days regardless of the initial diclofenac concentration.

The percentage of diclofenac removal in the abiotic culture obtained in the present study was lower compared to other studies where higher degradation percentages were obtained. However, this was due to the type of light used in this experiment. Generally, in photodegradation studies, compounds in solution

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are exposed to a source of UV light [52–54]. In addition to the type of light, Zhang et al. [52] established that the diclofenac photodegradation is reduced by the presence of nitrate ions and/or dissolved oxygen in the medium. Our studied culture medium had nitrate ions with an initial concentration of 2.94 mM.

4 Conclusions

Our results showed that diclofenac did not inhibit the growth of *N. oculata* CCAP 846/7 at the spiked concentration (16 µg/mL). The cultures of *N. oculata* CCAP 846/7 exposed to diclofenac presented an efficient removal of the drug (68.8%) in the medium, with the highest removal during the first 8 days. No statistically significant difference was found in diclofenac removal in cultures growing with complete medium and with different concentrations of nitrogen and carbon. *N. oculata* showed a low percentage of diclofenac adsorption at the cell wall (12.7%), suggesting that other processes are involved in the removal of the drug such as absorption. Abiotic cultures irradiated with light resulted in 23.6% of diclofenac degradation after 18 days. The results obtained point to the promissory use of *N. oculata* CCAP 846/7 as a bioremediation system for the removal of diclofenac; these results are expected to contribute to the knowledge on the application of microalgae for wastewater bioremediation, mainly on the removal of ECs such as pharmaceuticals.

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Declaration of Contributions: Sánchez-Sandoval and Navarro-Martínez performed the experimentation and drafted the manuscript. Sánchez-Sandoval, Navarro-Martínez, Castro-Tapia and González-Ortega participated in the analysis and interpretation of the data. Sánchez-Sandoval, González-Ortega and García-de la Cruz participated in the discussion of the results, reviewed and corrected the manuscript. Soria-Guerra supervised the experimentation, contributed to the drafted of the manuscript and reviewed the article for final approval.

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Appendix



Figure S1: Growth curves of cultures of *N. oculata* exposed to diclofenac (16 μ g/mL), under different concentrations of nitrogen (N) and carbon (C). The control culture did not contain diclofenac. Statistically significant differences with respect to the control are marked with an asterisk (ANOVA, p < 0.05, Dunnett test)

Basal Bold	1 L
Sodium nitrate	25 g
Magnesium sulfate heptahydrate	7.5 g
Sodium chloride	2.5 g
Potassium phosphate monohydrate	7.5 g
Potassium phosphate dihydrate	17.5 g
Metal solution	1 mL
$-ZnSO_4 - 7H_2O$	0.0088 g
- $MnCl_2 - 4H_2O$	0.0014 g
- <i>MoO</i> ₃	0.0007 g
- $CuSO_4 - 5H_2O$	0.0015 g
$-Co(NO_3)_2 - 6H_2O$	0.0004 g
$-H_3BO_3$	0.0114 g
-EDTA	0.05 g
-КОН	0.031 g
$-FeSO_4 - 7H_2O$	1 g
$-H_2SO_4$	1 mL

Table S1: Composition of the Bold nutrient solution

Note: The quantities indicated are those necessary to prepare 1 L of each solution and 1 mL in the case of the metal solution.